Section 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

PRODUCT NAME
Sodium azide

STATEMENT OF HAZARDOUS NATURE

NFPA

SUPPLIER
Company: Santa Cruz Biotechnology, Inc.
Address:
2145 Delaware Ave
Santa Cruz, CA 95060
Telephone: 800.457.3801 or 831.457.3800
Emergency Tel: CHEMWATCH: From within the US and Canada: 877-715-9305
Emergency Tel: From outside the US and Canada: +800 2436 2255 (1-800-CHEMCALL) or call +613 9573 3112

PRODUCT USE
Used for air bag inflation; preservative in diagnostic medicinals; intermediate in explosives manufacture; in weed and fruit rot control.

SYNONYMS
Na-N3, Azium, Kazoe, Smite, LC00001222

Section 2 - HAZARDS IDENTIFICATION

CANADIAN WHMIS SYMBOLS

EMERGENCY OVERVIEW
RISK
Heating may cause an explosion.
Contact with acids liberates very toxic gas.
Very toxic in contact with skin and if swallowed.
Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

POTENTIAL HEALTH EFFECTS
ACUTE HEALTH EFFECTS

SWALLOWED
- Severely toxic effects may result from the accidental ingestion of the material; animal experiments indicate that ingestion of
less than 5 gram may be fatal or may produce serious damage to the health of the individual.

- Azides, if swallowed, can cause breathlessness and rapid heart beat within 5 minutes. Nausea, vomiting, headache, restlessness and diarrhea can occur within 15 minutes. Other symptoms include low blood pressure which cannot be corrected, abnormal breathing, reduced body temperature, reduced blood pH, convulsions, collapse and death. Continued administration can cause increased sensitivity. Poisoning can cause headaches and acidosis. Several grams of sodium azide can cause liver, pulmonary and brain swelling with death occurring in less than an hour. Large doses of azide increases the blood pressure and causes generalized convulsions, followed by depression and collapse.

**EYE**
- Although the material is not thought to be an irritant, direct contact with the eye may cause transient discomfort characterized by tearing or conjunctival redness (as with windburn). Slight abrasive damage may also result. The material may produce foreign body irritation in certain individuals.

**SKIN**
- Skin contact with the material may produce severely toxic effects; systemic effects may result following absorption and these may be fatal.
- The material is not thought to be a skin irritant (as classified using animal models). Abrasive damage however, may result from prolonged exposures. Good hygiene practice requires that exposure be kept to a minimum and that suitable gloves be used in an occupational setting.
- Open cuts, abraded or irritated skin should not be exposed to this material.
- Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.
- Absorption by skin may readily exceed vapor inhalation exposure. Symptoms for skin absorption are the same as for inhalation.

**INHALED**
- Inhalation of dusts, generated by the material during the course of normal handling, may produce serious damage to the health of the individual.
- There is some evidence to suggest that the material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.
- Persons with impaired respiratory function, airway diseases and conditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled.
- Azide vapors are irritating and cause bronchitis and lung edema. If inhaled, sore throat, cough, dizziness, shortness of breath and fainting can result. Inhalation can result in similar symptoms as ingesting the substance. Other effects include eye irritation, headache, low blood pressure and collapse. Blindness, rigidity, liver and brain damage is possible.

**CHRONIC HEALTH EFFECTS**
- Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems.
- Long term exposure to high dust concentrations may cause changes in lung function i.e. pneumoconiosis; caused by particles less than 0.5 micron penetrating and remaining in the lung. Prime symptom is breathlessness; lung shadows show on X-ray. Prolonged exposures may be fatal. Continued exposure can cause increased sensitivity. Poisoning can cause headaches and acidosis. Several grams of sodium azide can cause liver, pulmonary and brain swelling with death occurring in less than an hour. Large doses of azide increases the blood pressure and causes generalized convulsions, followed by depression and collapse.

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**Section 3 - COMPOSITION / INFORMATION ON INGREDIENTS**

**HAZARD RATINGS**

<table>
<thead>
<tr>
<th>Hazard Class</th>
<th>Minimum Value</th>
<th>Maximum Value</th>
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</thead>
<tbody>
<tr>
<td>Flammability</td>
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<tr>
<td>Toxicity</td>
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<td>4</td>
</tr>
<tr>
<td>Body Contact</td>
<td>2</td>
<td>4</td>
</tr>
<tr>
<td>Reactivity</td>
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<td>4</td>
</tr>
<tr>
<td>Chronic</td>
<td>2</td>
<td>4</td>
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</tbody>
</table>

**NAME**

sodium azide

**CAS RN**

26628-22-8

**%**

> 99

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**Section 4 - FIRST AID MEASURES**

**SWALLOWED**

- **IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY.**
- Where Medical attention is not immediately available or where the patient is more than 15 minutes from a hospital or unless instructed otherwise:
  - For advice, contact a Poisons Information Center or a doctor.
  - Urgent hospital treatment is likely to be needed.
  - If conscious, give water to drink.
  - **INDUCE vomiting with fingers down the back of the throat, ONLY IF CONSCIOUS.** Lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. **NOTE:** Wear a protective glove when inducing vomiting by mechanical means.
  - In the mean time, qualified first-aid personnel should treat the patient following observation and employing supportive measures as indicated by the patient's condition.
  - If the services of a medical officer or medical doctor are readily available, the patient should be placed in his/her care and a copy of the MSDS should be provided. Further action will be the responsibility of the medical specialist.
  - If medical attention is not available on the worksite or surroundings send the patient to a hospital together with a copy of the MSDS.
  - Azide ingestions are potentially dangerous to health care providers. In the acid stomach, volatile and toxic hydrazoic acid is also produced in aqueous solutions of sodium azide, frequently complain about headache, Rapid falls in blood pressure can also result.
Isolate vomitus, gastric washings, dispose of azide residues promptly and safely. Keep patient in well ventilated area.

**EYE**
- If this product comes in contact with the eyes:
  - Immediately hold eyelids apart and flush the eye continuously with running water.
  - Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
  - Continue flushing until advised to stop by the Poisons Information Center or a doctor, or for at least 15 minutes.
  - Transport to hospital or doctor without delay.
- Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

**SKIN**
- If skin or hair contact occurs:
  - Immediately flush body and clothes with large amounts of water, using safety shower if available.
  - Quickly remove all contaminated clothing, including footwear.
  - Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Center.
  - Transport to hospital, or doctor.

**INHALED**
- If fumes or combustion products are inhaled remove from contaminated area.
- Lay patient down. Keep warm and rested.
- Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.
- Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.
- Transport to hospital, or doctor, without delay.

**NOTES TO PHYSICIAN**
- for poisons (where specific treatment regime is absent):

  **BASIC TREATMENT**
  - Establish a patent airway with suction where necessary.
  - Watch for signs of respiratory insufficiency and assist ventilation as necessary.
  - Administer oxygen by non-rebreather mask at 10 to 15 l/min.
  - Monitor and treat, where necessary, for pulmonary edema.
  - Monitor and treat, where necessary, for shock.
  - Anticipate seizures.
  - DO NOT use emetics. Where ingestion is suspected rinse mouth and give up to 200 ml water (5 ml/kg recommended) for dilution where patient is able to swallow, has a strong gag reflex and does not drool.

  **ADVANCED TREATMENT**
  - Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
  - Positive-pressure ventilation using a bag-valve mask might be of use.
  - Monitor and treat, where necessary, for arrhythmias.
  - Start an IV D5W TKO. If signs of hypovolemia are present use lactated Ringers solution. Fluid overload might create complications.
  - Drug therapy should be considered for pulmonary edema.
  - Hypotension with signs of hypovolemia requires the cautious administration of fluids. Fluid overload might create complications.
  - Treat seizures with diazepam.
  - Proparacaine hydrochloride should be used to assist eye irrigation.

**STORAGE**
- Store in a cool, dry place.
- Keep away from heat and direct sunlight.

**Hazardous waste**
- Properly dispose of used containers and unneeded supplies in accordance with local regulations.

**Section 5 - FIRE FIGHTING MEASURES**

<table>
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<tr>
<th>Property</th>
<th>Value</th>
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<td>Vapour Pressure (mmHG)</td>
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<tr>
<td>Upper Explosive Limit (%)</td>
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<tr>
<td>Specific Gravity (water=1)</td>
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</tr>
<tr>
<td>Lower Explosive Limit (%)</td>
<td>Not available.</td>
</tr>
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</table>

**EXTINGUISHING MEDIA**
- DO NOT use water.

**FIRE FIGHTING**
- Alert Emergency Responders and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves for fire only.
- Prevent, by any means available, spillage from entering drains or water course.
- Use fire fighting procedures suitable for surrounding area.
- Do not approach containers suspected to be hot.
- Cool fire exposed containers with water spray from a protected location.
If safe to do so, remove containers from path of fire.  
Equipment should be thoroughly decontaminated after use. 

GENERAL FIRE HAZARDS/HAZARDOUS COMBUSTIBLE PRODUCTS  
■ WARNING: May EXPLODE on heating!!!.  
■ Non combustible.  
■ Not considered to be a significant fire risk, however containers may burn.  
May emit poisonous fumes.  

FIRE INCOMPATIBILITY  
■ None known.  

PERSONAL PROTECTION  
Glasses:  
Full face- shield.  
Gloves:  
Respirator:  
Type B-P Filter of sufficient capacity  

Section 6 - ACCIDENTAL RELEASE MEASURES  

MINOR SPILLS  
■  
Clean up all spills immediately.  
Avoid contact with skin and eyes.  
Wear protective clothing, gloves, safety glasses and dust respirator.  
Sweep up or vacuum up (consider explosion-proof machines designed to be grounded during storage and use).  
Place in clean drum then flush area with water.  

MAJOR SPILLS  
■ Pollutant  
Clear area of personnel and move upwind.  
Alert Emergency Responders and tell them location and nature of hazard.  
May be violently or explosively reactive.  
Wear full body protective clothing with breathing apparatus.  
Prevent, by any means available, spillage from entering drains or water courses.  
Consider evacuation (or protect in place).  
No smoking, naked lights or ignition sources.  
Increase ventilation.  
Stop leak if safe to do so.  
Water spray or fog may be used to disperse / absorb vapour.  
Contain or absorb spill with sand, earth or vermiculite.  
Collect recoverable product into labelled containers for recycling.  
Collect solid residues and seal in labelled drums for disposal.  
Wash area and prevent runoff into drains.  
If contamination of drains or waterways occurs, advise emergency services.  

PROTECTIVE ACTIONS FOR SPILL  

FOOTNOTES  
1 PROTECTIVE ACTION ZONE is defined as the area in which people are at risk of harmful exposure. This zone assumes that random changes in wind direction confines the vapour plume to an area within 30 degrees on either side of the predominant wind direction, resulting in a crosswind protective action distance equal to the downwind protective action distance.  
2 PROTECTIVE ACTIONS should be initiated to the extent possible, beginning with those closest to the spill and working away from the site in the downwind direction. Within the protective action zone a level of vapour concentration may exist resulting in nearly all unprotected persons becoming incapacitated and unable to take protective action and/or incurring serious or irreversible health effects.  
3 INITIAL ISOLATION ZONE is determined as an area, including upwind of the incident, within which a high probability of localised wind reversal may expose nearly all persons without appropriate protection to life-threatening concentrations of the material.  
4 SMALL SPILLS involve a leaking package of 200 litres (55 US gallons) or less, such as a drum (jerrican or box with inner containers). Larger packages leaking less than 200 litres and compressed gas leaking from a small cylinder are also considered "small spills". LARGE SPILLS involve many small leaking packages or a leaking package of greater than 200 litres, such as a cargo tank, portable tank or a "one-tonne" compressed gas cylinder.  
6 IERG information is derived from CANUTEC - Transport Canada.
ACUTE EXPOSURE GUIDELINE LEVELS (AEGL) (in ppm)

AEGL 1: The airborne concentration of a substance above which it is predicted that the general population, including susceptible individuals, could experience notable discomfort, irritation, or certain asymptomatic nonsensory effects. However, the effects are not disabling and are transient and reversible upon cessation of exposure.

AEGL 2: The airborne concentration of a substance above which it is predicted that the general population, including susceptible individuals, could experience irreversible or other serious, long-lasting adverse health effects or an impaired ability to escape.

AEGL 3: The airborne concentration of a substance above which it is predicted that the general population, including susceptible individuals, could experience life-threatening health effects or death.

Section 7 - HANDLING AND STORAGE

PROCEDURE FOR HANDLING

- Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- Prevent concentration in hollows and sumps.
- DO NOT enter confined spaces until atmosphere has been checked.
- DO NOT allow material to contact humans, exposed food or food utensils.
- Avoid contact with incompatible materials.
- When handling, DO NOT eat, drink or smoke.
- Keep containers securely sealed when not in use.
- Avoid physical damage to containers.
- Always wash hands with soap and water after handling.
- Work clothes should be laundered separately.
- Launder contaminated clothing before re-use.
- Use good occupational work practice.
- Observe manufacturer's storing and handling recommendations.
- Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.

RECOMMENDED STORAGE METHODS

- Lined metal can, Lined metal pail/drum
- Plastic pail
- Polyliner drum
- Packing as recommended by manufacturer.
- Check all containers are clearly labeled and free from leaks.

For low viscosity materials
- Drums and jerricans must be of the non-removable head type.
- Where a can is to be used as an inner package, the can must have a screwed enclosure.

For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):
- Removable head packaging;
- Cans with friction closures and low pressure tubes and cartridges may be used.

- Where combination packages are used, and the inner packages are of glass, there must be sufficient inert cushioning material in contact with inner and outer packages * . - In addition, where inner packagings are glass and contain liquids of packing group I and II there must be sufficient inert absorbent to absorb any spillage *. - * unless the outer packaging is a close fitting molded plastic box and the substances are not incompatible with the plastic. All inner and sole packagings for substances that have been assigned to Packaging Groups I or II on the basis of inhalation toxicity criteria, must be hermetically sealed.

STORAGE REQUIREMENTS

- Store in original containers.
- Keep containers securely sealed.
- Store in a cool, dry, well-ventilated area.
- Store away from incompatible materials and foodstuff containers.
- Protect containers against physical damage and check regularly for leaks.
- Observe manufacturer's storing and handling recommendations.

SAFE STORAGE WITH OTHER CLASSIFIED CHEMICALS

X: Must not be stored together
O: May be stored together with specific preventions
+: May be stored together

X X + X X +
## Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

### Exposure Controls

<table>
<thead>
<tr>
<th>Source</th>
<th>Material</th>
<th>TWA ppm</th>
<th>TWA mg/m³</th>
<th>STEL ppm</th>
<th>STEL mg/m³</th>
<th>Peak ppm</th>
<th>Peak mg/m³</th>
<th>TWA F/CC</th>
<th>Notes</th>
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<td>Canada - British Columbia Occupational Exposure Limits</td>
<td>sodium azide (Sodium azide (as Sodium azide))</td>
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<td>US ACGIH Threshold Limit Values (TLV)</td>
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<td>US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants</td>
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<td>US - Hawaii Air Contaminant Limits</td>
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<tr>
<td>Canada - Prince Edward Island Occupational Exposure Limits</td>
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<tr>
<td>Canada - Prince Edward Island</td>
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TLV Basis: cardiac impairment; lung damage.
Occupational Exposure Limits

<table>
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<tr>
<th>Country/Region</th>
<th>Substance (CAS No)</th>
<th>TLV Basis</th>
<th>Maximum Exposure Limit</th>
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<tbody>
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<td>Canada - Quebec</td>
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<td>TLV Basis: Cardiac impairment; lung damage</td>
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<td>Canada - Nova Scotia</td>
<td>Sodium azide (Hydrazoic acid vapor)</td>
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</tbody>
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MATERIAL DATA

SODIUM AZIDE:
- It is the goal of the ACGIH (and other Agencies) to recommend TLVs (or their equivalent) for all substances for which there is evidence of health effects at airborne concentrations encountered in the workplace.
- At this time no TLV has been established, even though this material may produce adverse health effects (as evidenced in animal experiments or clinical experience). Airborne concentrations must be maintained as low as is practically possible and occupational exposure must be kept to a minimum.

NOTE: The ACGIH occupational exposure standard for Particles Not Otherwise Specified (P.N.O.S) does NOT apply.

Chemical irritants are chemicals that produce temporary and undesirable side-effects on the eyes, nose or throat. Historically occupational exposure standards for these irritants have been based on observation of workers’ responses to various airborne concentrations. Present day expectations require that nearly every individual should be protected against even minor sensory irritation and exposure standards are established using uncertainty factors or safety factors of 5 to 10 or more. On occasion animal no-observable-effect-levels (NOEL) are used to determine these limits where human results are unavailable. An additional approach, typically used by the TLV committee (USA) in determining respiratory standards for this group of chemicals, has been to assign ceiling values (TLV C) to rapidly acting irritants and to assign short-term exposure limits (TLV STELs) when the weight of evidence from irritation, bioaccumulation and other endpoints combine to warrant such a limit. In contrast the MAK Commission (Germany) uses a five-category system based on intensive odour, local irritation, and elimination half-life. However this system is being replaced to be consistent with the European Union (EU) Scientific Committee for Occupational Exposure Limits (SCOEL); this is more closely allied to that of the USA.

OSHA (USA) concluded that exposure to sensory irritants can:
- cause inflammation
- cause increased susceptibility to other irritants and infectious agents
- lead to permanent injury or dysfunction
- permit greater absorption of hazardous substances and
- acclimate the worker to the irritant warning properties of these substances thus increasing the risk of overexposure.

for sodium azide:
Deaths due to sodium azide exposure are due to acute cardiovascular collapse, and central respiratory paralysis. Inhalation (or ingestion) produces dizziness, weakness, blurred vision, slight dyspnea, tachypnea, hypotension, tachycardia, acidosis, abdominal pain and spasms. Serious exposures produce polydipsia, leukocytosis, pulmonary edema, bronchitis, convulsions, unconsciousness, and death. Solutions of sodium azide release hydrazoic acid which has the same degree of acute toxicity as the salt.

Exposures must be controlled to very low levels to avoid hypotension.

Estimation of a maximal intake of sodium azide over an 8-hour shift (assuming 10 m3 of air inhaled and a 75% retention) for a daily pulmonary absorption of 0.75 mg results in an ambient air concentration of 0.04 ppm (0.1 mg/m3). The permissible limits of exposure to both sodium and hydrogen azides are thought to incorporate a reasonable margin of safety against headache and other symptoms of systemic discomfort and are thought to be protective against significant changes in electrocardiogram readings and hypotension. An alert has been published by OSHA to highlight the fact that significant percutaneous absorption may occur and this route may contribute significantly to worker exposure.

PERSONAL PROTECTION

Consult your EHS staff for recommendations.

EYE
- Chemical goggles.
- Full face shield.
- Contact lenses pose a special hazard; soft contact lenses may absorb irritants and all lenses concentrate them.

HANDS/FEET
- Elbow length PVC gloves.
- Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include: such as:
- frequency and duration of contact.
chemical resistance of glove material,
glove thickness and
dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739).

- When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374) is recommended.
- When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374) is recommended.
- Contaminated gloves should be replaced.

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

OTHER

- Overalls.
- Eyewash unit.
- Barrier cream.
- Skin cleansing cream.

- Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.
- The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure - ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option).
- If available, Time Weighted Average (TWA) exposure limits, where they exist, will be useful in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.
- Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.
- Use approved positive flow mask if significant quantities of dust becomes airborne.
- Try to avoid creating dust conditions.

RESPIRATOR

<table>
<thead>
<tr>
<th>Protection Factor</th>
<th>Half-Face Respirator</th>
<th>Full-Face Respirator</th>
<th>Powered Air Respirator</th>
</tr>
</thead>
<tbody>
<tr>
<td>10 x PEL</td>
<td>B P1</td>
<td>-</td>
<td>B PAPR-P1</td>
</tr>
<tr>
<td>50 x PEL</td>
<td>Air-line*</td>
<td>-</td>
<td>Air-line*</td>
</tr>
<tr>
<td>100 x PEL</td>
<td>Air-line**</td>
<td>B P2 B P3</td>
<td>Air-line** B PAPR-P3</td>
</tr>
<tr>
<td>100+ x PEL</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
</tbody>
</table>

- Protection Factor: 10 x PEL, 50 x PEL, 100 x PEL, 100+ x PEL
- Half-Face Respirator: B P1, Air-line*
- Full-Face Respirator: B P2 B, Air-line*
- Powered Air Respirator: B PAPR-P1, B PAPR-P2, B PAPR-P3

- * - Negative pressure demand
- ** - Continuous flow

Explanation of Respirator Codes:

Class 1 low to medium absorption capacity filters.
Class 2 medium absorption capacity filters.
Class 3 high absorption capacity filters.

PAPR Powered Air Purifying Respirator (positive pressure) cartridge.

- Type A for use against certain organic gases and vapors.
- Type AX for use against low boiling point organic compounds (less than 65°C).
- Type B for use against certain inorganic gases and other acid gases and vapors.
- Type C for use against sulfur dioxide and other acid gases and vapors.
- Type K for use against ammonia and organic ammonia derivatives.

- Class P1 intended for use against mechanically generated particulates of sizes most commonly encountered in industry, e.g. asbestos, silica.
- Class P2 intended for use against both mechanically and thermally generated particulates, e.g. metal fume.
- Class P3 intended for use against all particulates containing highly toxic materials, e.g. beryllium.

The local concentration of material, quantity and conditions of use determine the type of personal protective equipment required.

Use appropriate NIOSH-certified respirator based on informed professional judgement. In conditions where no reasonable estimate of exposure can be made, assume the exposure is in a concentration IDLH and use NIOSH-certified full face pressure demand SCBA with a minimum service life of 30 minutes, or a combination full facepiece pressure demand SAR with auxiliary self-contained air supply. Respirators provided only for escape from IDLH atmospheres shall be NIOSH-certified for escape from the atmosphere in which they will be used.

ENGINEERING CONTROLS

- Local exhaust ventilation usually required. If risk of overexposure exists, wear an approved respirator. Correct fit is essential to obtain adequate protection an approved self contained breathing apparatus (SCBA) may be required in some situations.
- Provide adequate ventilation in warehouse or closed storage area.
- Provide ventilation for all operations where high concentrations of contaminants are generated. A typical ventilation system for a spray booth would include local exhaust ventilation to limit the concentration of airborne contaminants near the spray booth, and dilution ventilation to dilute the contaminants in the space outside the spray booth.

<table>
<thead>
<tr>
<th>Type of Contaminant</th>
<th>Air Speed:</th>
</tr>
</thead>
<tbody>
<tr>
<td>solvent, vapors, degreasing etc., evaporating from tank (in still air).</td>
<td>0.25-0.5 m/s (50-100 ft/min.)</td>
</tr>
<tr>
<td>aerosols, fumes from pouring operations, intermittent container filling, low speed conveyor transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)</td>
<td>0.5-1 m/s (100-200 ft/min.)</td>
</tr>
<tr>
<td>direct spray, spray painting in shallow booths, drum filling, conveyor loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)</td>
<td>1.2-5 m/s (200-500 ft/min.)</td>
</tr>
<tr>
<td>grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).</td>
<td>2.5-10 m/s (500-2000 ft/min.)</td>
</tr>
</tbody>
</table>
Within each range the appropriate value depends on:

<table>
<thead>
<tr>
<th>Lower end of the range</th>
<th>Upper end of the range</th>
</tr>
</thead>
<tbody>
<tr>
<td>1: Room air currents minimal or favorable to capture</td>
<td>1: Disturbing room air currents</td>
</tr>
<tr>
<td>2: Contaminants of low toxicity or of nuisance value only.</td>
<td>2: Contaminants of high toxicity</td>
</tr>
<tr>
<td>3: Intermittent, low production.</td>
<td>3: High production, heavy use</td>
</tr>
<tr>
<td>4: Large hood or large air mass in motion</td>
<td>4: Small hood-local control only</td>
</tr>
</tbody>
</table>

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 ft/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

### Section 9 - PHYSICAL AND CHEMICAL PROPERTIES

#### PHYSICAL PROPERTIES

- **Solid.**
- **Mixes with water.**
- **Contact with acids liberates very toxic gas.**

<table>
<thead>
<tr>
<th>State</th>
<th>Divided solid</th>
<th>Molecular Weight</th>
<th>65.02</th>
</tr>
</thead>
<tbody>
<tr>
<td>Melting Range (°F)</td>
<td>527 (decomposes)</td>
<td>Viscosity</td>
<td>Not Applicable</td>
</tr>
<tr>
<td>Boiling Range (°F)</td>
<td>Not applicable.</td>
<td>Solubility in water (g/L)</td>
<td>Miscible</td>
</tr>
<tr>
<td>Flash Point (°F)</td>
<td>Not available.</td>
<td>pH (1% solution)</td>
<td>Not available</td>
</tr>
<tr>
<td>Decomposition Temp (°F)</td>
<td>Not Available.</td>
<td>pH (as supplied)</td>
<td>Not applicable</td>
</tr>
<tr>
<td>Autoignition Temp (°F)</td>
<td>Not available.</td>
<td>Vapour Pressure (mmHg)</td>
<td>Not available.</td>
</tr>
<tr>
<td>Upper Explosive Limit (%)</td>
<td>Not available.</td>
<td>Specific Gravity (water=1)</td>
<td>1.85</td>
</tr>
<tr>
<td>Lower Explosive Limit (%)</td>
<td>Not available.</td>
<td>Relative Vapor Density (air=1)</td>
<td>2.2</td>
</tr>
<tr>
<td>Volatile Component (%vol)</td>
<td>Not available.</td>
<td>Evaporation Rate</td>
<td>Not available</td>
</tr>
</tbody>
</table>

#### APPEARANCE

Colourless hexagonal crystals with characteristic odour. Soluble in water and in liquid ammonia; hydrolyses in water to form hydrazoic acid. Slightly soluble in alcohol; insoluble in ether.

### Section 10 - CHEMICAL STABILITY

#### CONDITIONS CONTRIBUTING TO INSTABILITY

- Presence of heat source
- Presence of shock and friction
- Presence of incompatible materials.
- Product is considered stable.
- Hazardous polymerization will not occur.

#### STORAGE INCOMPATIBILITY

- Inorganic azides:
  - reacts with hot water
  - decompose explosively at elevated temperatures (above 275 C).
  - form ultra-sensitive explosive compounds with heavy metals, copper and its alloys, lead, silver and mercury, carbon disulfide, trifluoroacryloyl fluoride
  - react violently with acids, forming explosive hydrogen azide, with strong oxidisers, with bromine, barium carbonate, chromyl chloride, dimethyl sulfate, dibromomalonitrile
  - are incompatible with caustics, cyanuric chloride, benzoyl chloride, halogen acids, halogen compounds, metal oxides, metal sulfides, methyl azide, phosgene
- High nitrogen compounds are often unstable or explosive; the tendency is exaggerated by attachment of azide or diazonium groups, or a high-nitrogen heterocyclic nucleus.
- High-nitrogen chemical families include
  - azides
  - diazoazoles
  - diazonium salts
  - hydrazinium salts
  - N-nitro compounds
  - tetrazoles
  - triazoles
  - triazoles
  - Protect from light.

For incompatible materials - refer to Section 7 - Handling and Storage.
Section 11 - TOXICOLOGICAL INFORMATION

sodium azide

TOXICITY AND IRRITATION

■ unless otherwise specified data extracted from RTECS - Register of Toxic Effects of Chemical Substances.

<table>
<thead>
<tr>
<th>TOXICITY</th>
<th>IRRITATION</th>
</tr>
</thead>
<tbody>
<tr>
<td>Oral (man) LDLo: 143 mg/kg</td>
<td>Nil Reported</td>
</tr>
<tr>
<td>Oral (human) TDLo: 0.71 mg/kg</td>
<td></td>
</tr>
<tr>
<td>Oral (woman) TDLo: 3 mg/kg</td>
<td></td>
</tr>
<tr>
<td>Oral (rat) LD50: 27 mg/kg</td>
<td></td>
</tr>
</tbody>
</table>

General anaesthesia, somnolence, convulsions, headache, irritability, arrhythmias, dyspnea, respiratory stimulation, diarrhoea recorded.

CARCINOGEN

Sodium azide (as Hydrazoic acid vapor) US ACGIH Threshold Limit Values (TLV) - Carcinogens Carcinogen Category A4

SKIN

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Persistence: Water/Soil</th>
<th>Persistence: Air</th>
<th>Bioaccumulation</th>
<th>Mobility</th>
</tr>
</thead>
<tbody>
<tr>
<td>sodium azide</td>
<td>LOW</td>
<td>LOW</td>
<td></td>
<td>HIGH</td>
</tr>
</tbody>
</table>

Section 12 - ECOLOGICAL INFORMATION

Refer to data for ingredients, which follows:

SODIUM AZIDE:

■ Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

■ Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

■ DO NOT discharge into sewer or waterways.

Toxicity Fish: 24hr LC50: 5000 ppm

Toxicity Fish (L. macrochirus) 24hr LC50: 1.5 ppm *

Acute toxicity for lower organisms: 5 mg/l approx. *

Acute toxicity for cold-blooded animals: 1 mg/l *

Microorganisms LC50: 5 mg/l * [Merck]

Ecotoxicity

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Persistence: Water/Soil</th>
<th>Persistence: Air</th>
<th>Bioaccumulation</th>
<th>Mobility</th>
</tr>
</thead>
<tbody>
<tr>
<td>sodium azide</td>
<td>LOW</td>
<td>LOW</td>
<td></td>
<td>HIGH</td>
</tr>
</tbody>
</table>

Section 13 - DISPOSAL CONSIDERATIONS

US EPA Waste Number & Descriptions

A. General Product Information

Reactivity characteristic: use EPA hazardous waste number D003 (waste code R).

B. Component Waste Numbers

When sodium azide is present as a solid waste as a discarded commercial chemical product, off-specification species, as a container residue, or a spill residue, use EPA waste number P105 (waste code T).

Disposal Instructions

All waste must be handled in accordance with local, state and federal regulations.

Puncture containers to prevent re-use and bury at an authorized landfill.

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:
Reduction
Reuse
Recycling
Disposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.

DO NOT allow wash water from cleaning equipment to enter drains. Collect all wash water for treatment before disposal.

DO NOT allow wash water from cleaning equipment to enter drains. Collect all wash water for treatment before disposal.

NOTE: If azides are poured down the drain they may accumulate in traps and react with lead and copper plumbing to form highly explosive compounds which may be detonated, inadvertently, by plumbers. On disposal, flush with large volumes of water to prevent azide build-up.

Recycle wherever possible or consult manufacturer for recycling options.

Consult Waste Management Authority for disposal.

Bury residue in an authorized landfill.

Recycle containers wherever possible, or dispose of in an authorized landfill.

---

**Section 14 - TRANSPORTATION INFORMATION**

DOT:

- Symbols: None
- Hazard class or Division: 6.1
- Identification Numbers: UN1687, PG: II
- Label Codes: 6.1
- Special provisions: IB8, IP2, IP4
- Packaging: Exceptions: 153
- Packaging: Non-bulk: 212
- Packaging: Exceptions: 153
- Quantity limitations: 25 kg
- Quantity Limitations: Cargo aircraft only: 100 kg
- Vessel stowage: Location: A
- Vessel stowage: Other: 36, 52, 91

**Air Transport IATA:**

- ICAO/IATA Class: 6.1
- ICAO/IATA Subrisk: None
- UN/ID Number: 1687
- Packing Group: II
- Special provisions: None

**Shipping Name:** SODIUM AZIDE

**Maritime Transport IMDG:**

- IMDG Class: 6.1
- IMDG Subrisk: None
- UN Number: 1687
- Packing Group: II
- EMS Number: F-A,S-A
- Special provisions: None
- Limited Quantities: 500 g

**Shipping Name:** SODIUM AZIDE

---

**Section 15 - REGULATORY INFORMATION**

sodium azide (CAS: 26628-22-8) is found on the following regulatory lists;

- Canada - Alberta Occupational Exposure Limits
- Canada - British Columbia Occupational Exposure Limits
- Canada - Northwest Territories Occupational Exposure Limits (English)
- Canada - Nova Scotia Occupational Exposure Limits
- Canada - Ontario Occupational Exposure Limits
- Canada - Prince Edward Island Occupational Exposure Limits
- Canada - Quebec Permissible Exposure Values for Airborne Contaminants (English)
- Canada - Saskatchewan Industrial Hazardous Substances
- Canada - Saskatchewan Occupational Health and Safety Regulations - Contamination Limits
- Canada - Yukon Permissible Concentrations for Airborne Contaminant Substances
- Canada Domestic Substances List (DSL)
- Canada Ingredient Disclosure List (SOR/88-64)
- Canada Toxicological Index Service - Workplace Hazardous Materials Information System - WHMIS (English)
- Canada Toxicological Index Service - Workplace Hazardous Materials Information System - WHMIS (French)
- OECD Representative List of High Production Volume (HPV) Chemicals
- US - Alaska Limits for Air Contaminants
- US - California Occupational Safety and Health Regulations (CAL/OSHA) - Hazardous Substances List
- US - California Permissible Exposure Limits for Chemical Contaminants
- US - Connecticut Hazardous Air Pollutants
- US - Hawaii Air Contaminant Limits
- US - Massachusetts Oil & Hazardous Material List
- US - Michigan Exposure Limits for Air Contaminants
- US - Minnesota Hazardous Substance List
- US - Minnesota Permissible Exposure Limits (PELs)
- US - New Jersey Right to Know Hazardous Substances
- US - Pennsylvania - Hazardous Substance List
- US - Rhode Island Hazardous Substance List
- US - Tennessee Occupational Exposure Limits
- US - Vermont Hazardous Waste - Acutely Hazardous Wastes
- US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
- US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants
- US - Washington Discarded Chemical Products List
- "P" Chemical Products
- US - Washington Permissible exposure limits of air contaminants
- US ACGIH Threshold Limit Values (TLV)
Section 16 - OTHER INFORMATION

LIMITED EVIDENCE

- Inhalation may produce serious health damage*.
- Cumulative effects may result following exposure*.
- May produce discomfort of the respiratory system*.
* (limited evidence).

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- Classification of the mixture and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.
- A list of reference resources used to assist the committee may be found at: www.chemwatch.net/references.
- The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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