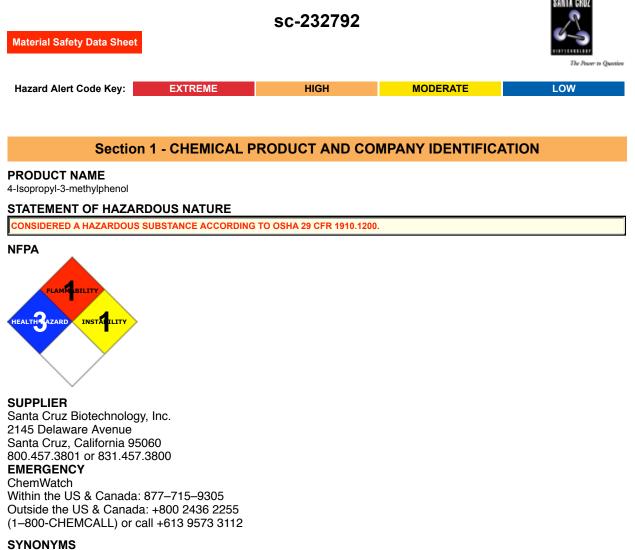
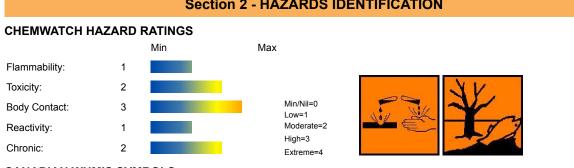
# 4-Isopropyl-3-methylphenol



C10-H14-O, (CH3)2CHC6H3(CH3)OH, o-cymen-5-ol, 3-methyl-4-isopropylphenol, 3-methyl-4-(1-methylethyl)phenol, "phenol, 3-methyl-4-isopropylphenol, 3-methyl-3-me 4-(1-methylethyl)-", Biosol



### Section 2 - HAZARDS IDENTIFICATION

### **CANADIAN WHMIS SYMBOLS**



## EMERGENCY OVERVIEW

Causes burns. Risk of serious damage to eyes. Toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

### POTENTIAL HEALTH EFFECTS

### ACUTE HEALTH EFFECTS

### SWALLOWED

The material can produce chemical burns within the oral cavity and gastrointestinal tract following ingestion.

Accidental ingestion of the material may be damaging to the health of the individual.

Some phenol derivatives can cause damage to the digestive system.

If absorbed, profuse sweating, thirst, nausea, vomiting diarrhea, cyanosis, restlessness, stupor, low blood pressure, gasping, abdominal pain, anemia, convulsions, coma and lung swelling can happen followed by pneumonia.

Thymol resembles phenol in its systemic actions but less toxic due to its lower solubility.

Ingestion may cause burning pain in the oesophagus, nausea, abdominal pain, vomiting, dizziness, convulsions, coma, cyanosis, central hyperactivity (e.

### EYE

The material can produce chemical burns to the eye following direct contact.

Vapors or mists may be extremely irritating.

■ If applied to the eyes, this material causes severe eye damage.

Some phenol derivatives may produce mild to severe eye irritation with redness, pain and blurred vision.

Permanent eye injury may occur; recovery may also be complete or partial.

Direct eye contact with acid corrosives may produce pain, tears, sensitivity to light and burns.

Mild burns of the epithelia generally recover rapidly and completely.

### SKIN

The material can produce chemical burns following direct contact with the skin.

Skin contact with the material may damage the health of the individual; systemic effects may result following absorption.

Phenol and its derivatives can cause severe skin irritation if contact is maintained, and can be absorbed to the skin affecting the cardiovascular and central nervous system.

Effects include sweating, intense thirst, nausea and vomiting, diarrhea, cyanosis, restlessness, stupor, low blood pressure, hyperventilation, abdominal pain, anemia, convulsions, coma, lung swelling followed by pneumonia.

Open cuts, abraded or irritated skin should not be exposed to this material.

Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects.

Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

Skin contact with acidic corrosives may result in pain and burns; these may be deep with distinct edges and may heal slowly with the formation of scar tissue.

### INHALED

■ If inhaled, this material can irritate the throat andlungs of some persons.

■ Inhalation of dusts, generated by the material during the course of normal handling, may be damaging to the health of the individual.

Persons with impaired respiratory function, airway diseases and conditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled.

Corrosive acids can cause irritation of the respiratory tract, with coughing, choking and mucous membrane damage.

There may be dizziness, headache, nausea and weakness.

If phenols are absorbed via the lungs, systemic effects may occur affecting the cardiovascular and nervous systems.

Inhalation can result in profuse perspiration, intense thirst, nausea, vomiting, diarrhea, cyanosis, restlessness, stupor, falling blood pressure, hyperventilation, abdominal pain, anemia, convulsions, coma, swelling and inflammation of the lung.

### CHRONIC HEALTH EFFECTS

■ Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems.

Exposure to the material may cause concerns for human fertility, on the basis that similar materials provide some evidence of impaired fertility in the absence of toxic effects, or evidence of impaired fertility occurring at around the same dose levels as other toxic effects, but which are not a secondary non-specific consequence of other toxic effects.

Long-term exposure to phenol derivatives can cause skin inflammation, loss of appetite and weight, weakness, muscle aches and pain, liver damage, dark urine, loss of nails, skin eruptions, diarrhea, nervous disorders with headache, salivation, fainting, discoloration of the skin and eyes, vertigo and mental disorders, and damage to the liver and kidneys.

Repeated or prolonged exposure to acids may result in the erosion of teeth, swelling and or ulceration of mouth lining. Irritation of airways to lung, with cough, and inflammation of lung tissue often occurs.

Exposure to alkyl phenolics is associated with reduced sperm count andfertility in males.

	Section 3 - COMPOSITION / INFORMATION ON INGREDIENTS						
NAME	с	CAS RN	%				
p-thymol	3:	228-02-2	>98				

### **Section 4 - FIRST AID MEASURES**

### SWALLOWED

· For advice, contact a Poisons Information Center or a doctor at once. · Urgent hospital treatment is likely to be needed.

### EYE

■ If this product comes in contact with the eyes: • Immediately hold eyelids apart and flush the eye continuously with running water. • Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.

### SKIN

■ If skin or hair contact occurs: · Immediately flush body and clothes with large amounts of water, using safety shower if available. · Quickly remove all contaminated clothing, including footwear.

#### INHALED

· If fumes or combustion products are inhaled remove from contaminated area. · Lay patient down. Keep warm and rested. Inhalation of vapors or aerosols (mists, fumes) may cause lung edema. Corrosive substances may cause lung damage (e.g.

### NOTES TO PHYSICIAN

■ For acute or short term repeated exposures to phenols/ cresols:

· Phenol is absorbed rapidly through lungs and skin. [Massive skin contact may result in collapse and death]\*

• [Ingestion may result in ulceration of upper respiratory tract; perforation of esophagus and/or stomach, with attendant complications, may occur. Esophageal stricture may occur.]\*.

Section 5 - FIRE FIGHTING MEASURES					
Vapour Pressure (mmHG):	Not available				
Upper Explosive Limit (%):	Not available				
Specific Gravity (water=1):	Not available				
Lower Explosive Limit (%):	Not available				

### **EXTINGUISHING MEDIA**

· Foam.

### · Dry chemical powder.

#### FIRE FIGHTING

· Alert Emergency Responders and tell them location and nature of hazard.

· Wear full body protective clothing with breathing apparatus.

consider evacuation by 800 metres in all directions.

When any large container (including road and rail tankers) is involved in a fire,

### GENERAL FIRE HAZARDS/HAZARDOUS COMBUSTIBLE PRODUCTS

· Combustible solid which burns but propagates flame with difficulty.

Avoid generating dust, particularly clouds of dust in a confined or unventilated space as dusts may form an explosive mixture with air, and any source of ignition, i.e. flame or spark, will cause fire or explosion. Dust clouds generated by the fine grinding of the solid are a particular hazard; accumulations of fine dust may burn rapidly and fiercely if ignited.

Combustion products include: carbon monoxide (CO), carbon dioxide (CO2), other pyrolysis products typical of burning organic material. May emit corrosive fumes.

### FIRE INCOMPATIBILITY

Avoid contamination with oxidizing agents i.e. nitrates, oxidizing acids, chlorine bleaches, pool chlorine etc. as ignition may result.

### PERSONAL PROTECTION

Glasses: Safety Glasses. Full face- shield. Gloves: Respirator: Particulate dust filter.

### Section 6 - ACCIDENTAL RELEASE MEASURES

#### MINOR SPILLS

- · Remove all ignition sources.
- · Clean up all spills immediately.
- $\cdot$  Avoid contact with skin and eyes.
- $\cdot$  Control personal contact by using protective equipment.
- $\cdot$  Use dry clean up procedures and avoid generating dust.
- $\cdot$  Place in a suitable, labelled container for waste disposal.
- MAJOR SPILLS
- · Clear area of personnel and move upwind.
- · Alert Emergency Responders and tell them location and nature of hazard.

### Section 7 - HANDLING AND STORAGE

### **PROCEDURE FOR HANDLING**

· Avoid all personal contact, including inhalation.

· Wear protective clothing when risk of exposure occurs.

Empty containers may contain residual dust which has the potential to accumulate following settling. Such dusts may explode in the presence of an appropriate ignition source.

· Do NOT cut, drill, grind or weld such containers.

· In addition ensure such activity is not performed near full, partially empty or empty containers without appropriate workplace safety authorisation or permit.

### **RECOMMENDED STORAGE METHODS**

- DO NOT use aluminum or galvanized containers.
- · Lined metal can, Lined metal pail/drum
- · Plastic pail.
- For low viscosity materials
- $\cdot$  Drums and jerricans must be of the non-removable head type.
- · Where a can is to be used as an inner package, the can must have a screwed enclosure.

### STORAGE REQUIREMENTS

· Store in original containers.

· Keep containers securely sealed.

### Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

### **EXPOSURE CONTROLS**

Source	Material	TWA ppm	TWA mg/m³	STEL ppm	STEL mg/m³	Peak ppm	Peak mg/m³	TWA F/CC	Notes
Canada - British Columbia Occupational Exposure Limits	p-thymol (Particles (Insoluble or Poorly Soluble) Not Otherwise Classified (PNOC))		10 (N)						
US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants	p-thymol (Particulates not otherwise regulated (PNOR)(f)- Respirable fraction)		5						
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	p-thymol (Particulates not otherwise regulated Respirable fraction)		5						
US - California Permissible Exposure Limits for Chemical Contaminants	p-thymol (Particulates not otherwise regulated Respirable fraction)		5						(n)
US - Oregon Permissible Exposure Limits (Z-1)	p-thymol (Particulates not otherwise regulated (PNOR) (f) Total Dust)	-	10						Bold print identifies substances for which the Oregon Permissible Exposure Limits (PELs) are different than the federal Limits. PNOR means "particles not otherwise regulated."

US - Michigan Exposure Limits for Air Contaminants	p-thymol (Particulates not otherwise regulated, Respirable dust)	5	
Canada - Prince Edward Island Occupational Exposure Limits	p-thymol (Particles (Insoluble or Poorly Soluble) [NOS] Inhalable particles)	10	See Appendix B current TLV/BEI Book
US - Oregon Permissible Exposure Limits (Z-1)	p-thymol (Particulates not otherwise regulated - (PNOR) (f) Respirable Fraction)	5	Bold print identifies substances for which the Oregon Permissible Exposure Limits (PELs) are different than the federal Limits. PNOR means "particles not otherwise regulated."
ENDOELTABLE			

### PERSONAL PROTECTION



### RESPIRATOR

BR2 EYE

· Chemical goggles.

· Full face shield.

### HANDS/FEET

■ Wear chemical protective gloves, eg. PVC.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include: such as:

· frequency and duration of contact,

chemical resistance of glove material,

· glove thickness and

· dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739).

• When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374) is recommended.

· When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374) is recommended.

· Contaminated gloves should be replaced.

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

· Neoprene gloves.

### OTHER

· Overalls.

· PVC Apron.

#### **ENGINEERING CONTROLS**

· Local exhaust ventilation is required where solids are handled as powders or crystals; even when particulates are relatively large, a certain proportion will be powdered by mutual friction.

· Exhaust ventilation should be designed to prevent accumulation and recirculation of particulates in the workplace.

### Section 9 - PHYSICAL AND CHEMICAL PROPERTIES

### PHYSICAL PROPERTIES

Does not mix with water. Corrosive.			
State	DIVIDED SOLID	Molecular Weight	150.22
Melting Range (°F)	232- 237	Viscosity	Not Applicable
Boiling Range (°F)	Not available	Solubility in water (g/L)	Partly miscible
Flash Point (°F)	Not available	pH (1% solution)	Not available
Decomposition Temp (°F)	Not Available	pH (as supplied)	Not applicable
Autoignition Temp (°F)	Not available	Vapour Pressure (mmHG)	Not available
Upper Explosive Limit (%)	Not available	Specific Gravity (water=1)	Not available
Lower Explosive Limit (%)	Not available	Relative Vapor Density (air=1)	>1
Volatile Component (%vol)	Not available	Evaporation Rate	Not available

#### APPEARANCE

White crystalline solid; does not mix well with water. Soluble in alkali hydroxides.

Environmental toxicity is a function of the n-octanol/ water partition coefficient (log Pow, log Kow). Phenols with log Pow >7.4 are expected to exhibit low toxicity to aquatic organisms.<\div>.

Material

Value

### Section 10 - CHEMICAL STABILITY

#### CONDITIONS CONTRIBUTING TO INSTABILITY

- · Presence of incompatible materials.
- · Product is considered stable.

### STORAGE INCOMPATIBILITY

For thymol:

Avoid storage, acetanilide, antipyrine, camphor, choral hydrate, menthol, quinine sulfate, salol and urethane.

Reacts with mild steel, galvanized steel / zinc producing hydrogen gas which may form an explosive mixture with air.

• Phenols are incompatible with strong reducing substances such as hydrides, nitrides, alkali metals, and sulfides.

· Avoid use of aluminium, copper and brass alloys in storage and process equipment.

· Heat is generated by the acid-base reaction between phenols and bases.

Phenols are sulfonated very readily (for example, by concentrated sulfuric acid at room temperature), these reactions generate heat.

- Phenols are nitrated very rapidly, even by dilute nitric acid.
- · Nitrated phenols often explode when heated. Many of them form metal salts that tend toward detonation by rather mild shock.
- · Avoid oxidizing agents, acids, acid chlorides, acid anhydrides.

For incompatible materials - refer to Section 7 - Handling and Storage.

### Section 11 - TOXICOLOGICAL INFORMATION

p-thymol

### TOXICITY AND IRRITATION

P-THYMOL:

unless otherwise specified data extracted from RTECS - Register of Toxic Effects of Chemical Substances.

TOXICITY IRRITATION

Oral (mouse) LD50: 6280 mg/kg Nil Reported

Subcutaneous (mouse) LD50: 184 mg/mg

The material may be irritating to the eye, with prolonged contact causing inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis.

Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production.

The material may produce respiratory tract irritation, and result in damage to the lung including reduced lung function.

The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.

Thymol is found in the naturally occurring herb Thyme (Thymus vulgaris). Thyme is used as a food seasoning ingredient, and is generally recognized as a safe (GRAS) natural seasoning by the U.S. Food and Drug Administration (FDA). Thyme oil also is recognized as a GRAS essential oil by FDA. As a result, a large numbers of humans have been and continue to be regularly exposed to the active ingredient via

physical contact and in their diet with no known reports of adverse effects. Subchronic, chronic, immune, endocrine, or non-dietary exposure issues, as they may affect children and the general population, have not been identified. Thymol is a constituent of a mixture of organic compounds known to be rapidly degraded in the environment to elemental compounds

Thymol is added to the anaesthetic halothane as a preservative (0.01%) and is considered inactive (by FDA. 1997) at this concentration. Primary eye irritation Thymol instilled into the eyes of rabbits caused corneal opacity which persisted to the final observation at day 28. Conjunctival irritation persisted to day 14, and iritis resolved by day 7. The test material is a severe eye irritant Primary dermal irritation - Thymol is corrosive to the skin of rabbits.

Mutagenicity: Thymol has been reported to be non-mutagenic in multiple Ames tests (strains TA97, TA98, and TA100 w and w/out metabolic transformation with S9 incubation but positive in unscheduled DNA synthesis (liquid scintillation), sister chromatid exchange, and cell transformation tests in Syrian hamster embryo cells in culture In addition, thymol does not induce chromosomal aberrations in Allium cepa (Grant, 1982). Steam distilled extracts of three species of Thymus (capitatus, citriodorus, vulgaris) also were negative for DNA damaging activity and mutagenicity in the Ames test (strains TA1535, TA1537, TA98, and TA100 with and w/out metabolic activation). They were also non-mutagenic in a salmonella/microsome assay, did not induce the formation of micronuclei in mice, even when orally dosed in the toxic range (1100 mg/kg bw). Further, in the A/He strain of mice, thymol did not increase the incidence of spontaneous lung tumors following repeated intraperitoneal dosing . Overall, the weight of evidence suggests that thymol is not genotoxic or mutagenic. Endocrine systems: Thymol is not a known endocrine disruptor nor is it related to any class of known endocrine disruptors.

# Section 12 - ECOLOGICAL INFORMATION

Toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment. This material and its container must be disposed of as hazardous waste. Avoid release to the environment.

Refer to special instructions/ safety data sheets.

#### Ecotoxicity

Ingredient	Persistence: Water/Soil	Persistence: Air	Bioaccumulation	Mobility
p-thymol	HIGH		LOW	MED

### Section 13 - DISPOSAL CONSIDERATIONS

### **US EPA Waste Number & Descriptions**

A. General Product Information

Corrosivity characteristic: use EPA hazardous waste number D002 (waste code C)

#### Disposal Instructions

All waste must be handled in accordance with local, state and federal regulations.

| Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- · Reduction
- · Reuse
- · Recycling

· Disposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.

DO NOT allow wash water from cleaning equipment to enter drains. Collect all wash water for treatment before disposal.

· Recycle wherever possible.

· Consult manufacturer for recycling options or consult Waste Management Authority for disposal if no suitable treatment or disposal facility can be identified.

### **Section 14 - TRANSPORTATION INFORMATION**



DOT: Symbols: None Hazard class or Division: 8 Identification Numbers: UN3261 PG: III Label Codes: 8 Special provisions: IB8, IP3, T1, TP33 Packaging: Exceptions: 154 Packaging: Non- bulk: 213 Packaging: Exceptions: 154 Quantity limitations: 25 kg Passenger aircraft/rail: Quantity Limitations: Cargo 100 kg Vessel stowage: Location: A aircraft only: Vessel stowage: Other: None Hazardous materials descriptions and proper shipping names: Corrosive solid, acidic, organic, n.o.s. Air Transport IATA: ICAO/IATA Class: 8 ICAO/IATA Subrisk: None UN/ID Number: 3261 Packing Group: III Special provisions: A3 Cargo Only Packing Instructions: 100 kg Maximum Qty/Pack: 25 kg Passenger and Cargo Passenger and Cargo Packing Instructions: 864 Maximum Qty/Pack: 860 Passenger and Cargo Limited Quantity Passenger and Cargo Limited Quantity Packing Instructions: 5 kg Maximum Qty/Pack: Y845 Shipping Name: CORROSIVE SOLID, ACIDIC, ORGANIC, N.O.S. \*(CONTAINS P-THYMOL) Maritime Transport IMDG: IMDG Class: 8 IMDG Subrisk: None UN Number: 3261 Packing Group: III EMS Number: F-A, S-B Special provisions: 223 274

Limited Quantities: 5 kg Marine Pollutant: Yes Shipping Name: CORROSIVE SOLID, ACIDIC, ORGANIC, N.O.S. (contains p-thymol)

### **Section 15 - REGULATORY INFORMATION**

### p-thymol (CAS: 3228-02-2) is found on the following regulatory lists;

"Canada Domestic Substances List (DSL)","US Cosmetic Ingredient Review (CIR) Cosmetic ingredients found safe, with qualifications"

### Section 16 - OTHER INFORMATION

#### ND

Substance CAS Suggested codes p- thymol 3228- 02- 2 R43 N; R50/53

Reasonable care has been taken in the preparation of this information, but the author makes no warranty of merchantability or any other warranty, expressed or implied, with respect to this information. The author makes no representations and assumes no liability for any direct, incidental or consequential damages resulting from its use. For additional technical information please call our toxicology department on +800 CHEMCALL.

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references. A list of reference resources used to assist the committee may be found at: www.chemwatch.net/references.

■ The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

This document is copyright. Apart from any fair dealing for the purposes of private study, research, review or criticism, as permitted under the Copyright Act, no part may be reproduced by any process without written permission from CHEMWATCH. TEL (+61 3) 9572 4700.

Issue Date: Oct-5-2008 Print Date:Apr-7-2011