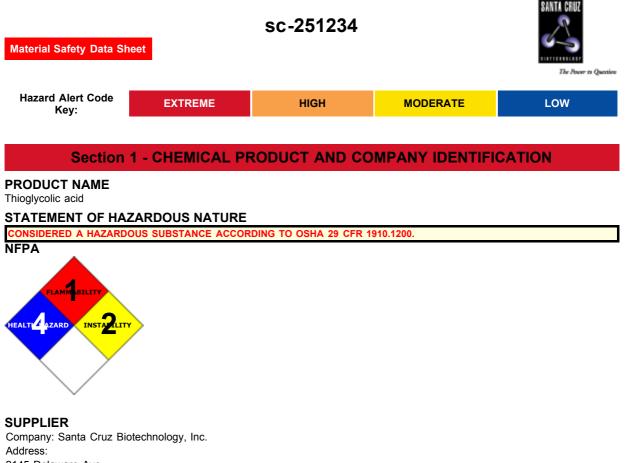
# Thioglycolic acid



Address: 2145 Delaware Ave Santa Cruz, CA 95060 Telephone: 800.457.3801 or 831.457.3800 Emergency Tel: CHEMWATCH: From within the US and Canada: 877-715-9305 Emergency Tel: From outside the US and Canada: +800 2436 2255 (1-800-CHEMCALL) or call +613 9573 3112

#### **PRODUCT USE**

Used as a sensitive reagent for iron, molybdenum, silver, tin. Used in the manufacture of thioglycolates. The ammonium and sodium salts are commonly used for cold waving and the calcium salt is a depilatory. Used in the manufacture of pharmaceuticals.

## **SYNONYMS**

C2-H4-O2-S, HSCH2COOH, "2-thioglycolic acid", "2-thioglycolic acid", mercaptoacetate, "alpha-mercaptoacetic acid", "2-mercaptoacetic acid", "2-mercaptoacetic acid", "2-mercaptoacetic acid", "thioglycollic acid", "thiovanic acid", "glycolic acid, thio-", "glycolic acid, 2-thio-", "glycolic acid, 2-thio-"

# Section 2 - HAZARDS IDENTIFICATION

## **CANADIAN WHMIS SYMBOLS**



#### EMERGENCY OVERVIEW RISK

Contact with acids liberates very toxic gas. Causes burns. Risk of serious damage to eyes. Toxic by inhalation, in contact with skin and if swallowed.

POTENTIAL HEALTH EFFECTS ACUTE HEALTH EFFECTS

#### SWALLOWED

The material can produce chemical burns within the oral cavity and gastrointestinal tract following ingestion.

Toxic effects may result from the accidental ingestion of the material; animal experiments indicate that ingestion of less than 40 gram may be fatal or may produce serious damage to the health of the individual.

The material can produce severe chemical burns within the oral cavity and gastrointestinal tract following ingestion.

Thioglycolate salts may produce decreased blood sugar levels, central nervous system depression, labored breathing, and convulsions.

• Symptoms of exposure to thioglycolic acid include weakness, shortness of breath and liver damage. It has been asserted that the acid releases significant amounts of hydrogen sulfide in contact with gastric juices.

Thioglycolic acid exhibits similar toxicology to acetic acid and is more injurious than concentrated mineral acids of the same pH. In a single oral dose of a 10% aqueous solutions, given by intubation to female rats, death was first observed at 125 mg/kg. Necroscopy indicated hepatic involvement.

#### EYE

• The material can produce chemical burns to the eye following direct contact. Vapors or mists may be extremely irritating.

■ The material can produce severe chemical burns to the eye following direct contact. Vapors or mists may be extremely irritating.

If applied to the eyes, this material causes severe eye damage.

■ Instillation of thioglycolic acid into rabbit eye resulted in severe pain, severe conjunctival inflammation, dense corneal opacity and severe iritis. After 14-days the prognosis had not improved even when the eyes were washed immediately following the application.

#### SKIN

The material can produce chemical burns following direct contactwith the skin.

Skin contact with the material may produce toxic effects; systemic effectsmay result following absorption.

The material can produce severe chemical burns following direct contactwith the skin.

■ Fatalities were produced by topical application of a 10% solution of thioglycolic acid to guinea pigs at less than 5 ml/kg. Signs of intoxication included weakness, gasping and convulsions. Professional hair-dressers exposed to thioglycolate products show skin irritation and skin sensitisation.

Open cuts, abraded or irritated skin should not be exposed to this material.

• Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

The material is a vesicant causing blistering on contact.

#### INHALED

■ Inhalation of vapors or aerosols (mists, fumes), generated by the material during the course of normal handling, may produce toxic effects.

The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.

Inhalation of thioglycolic acid mists may cause olfactory paralysis, weakness, shortness of breath and liver damage.

#### **CHRONIC HEALTH EFFECTS**

Repeated or prolonged exposure to corrosives may result in the erosion of teeth, inflammatory and ulcerative changes in the mouth and necrosis (rarely) of the jaw. Bronchial irritation, with cough, and frequent attacks of bronchial pneumonia may ensue. Gastrointestinal disturbances may also occur. Chronic exposures may result in dermatitis and/or conjunctivitis.

Long-term exposure to respiratory irritants may result in disease of the airways involving difficult breathing and related systemic problems.

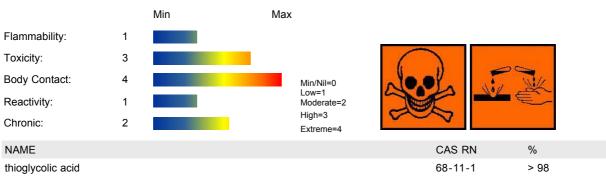
Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems.

There is limited evidence that, skin contact with this product is more likely to cause a sensitization reaction in some persons compared to the general population.

Chronic occupational exposure to thioglycolate salts has produced allergic reactions such as, collection of fluid under the skin, burning of the skin, reddening and hemorrhage under skin surface, eczema like dermatitis of the scalp or hands and bleeding under the skin.

## Section 3 - COMPOSITION / INFORMATION ON INGREDIENTS

#### **HAZARD RATINGS**



# **Section 4 - FIRST AID MEASURES**

## SWALLOWED

- For advice, contact a Poisons Information Center or a doctor at once.
- Urgent hospital treatment is likely to be needed.
- If swallowed do NOT induce vomiting.
- If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and

prevent aspiration.

- Observe the patient carefully.
- Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.
- Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.
- Transport to hospital or doctor without delay.

EYE

- If this product comes in contact with the eyes:
- Immediately hold eyelids apart and flush the eye continuously with running water.
- Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
- Continue flushing until advised to stop by the Poisons Information Center or a doctor, or for at least 15 minutes.
- Transport to hospital or doctor without delay.
- Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

SKIN

- If skin or hair contact occurs:
- Immediately flush body and clothes with large amounts of water, using safety shower if available.
- Quickly remove all contaminated clothing, including footwear.
- Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Center.

• Transport to hospital, or doctor.

INHALED

- If fumes or combustion products are inhaled remove from contaminated area.
- Lay patient down. Keep warm and rested.
- Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.
- Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.
- Transport to hospital, or doctor, without delay.

Inhalation of vapors or aerosols (mists, fumes) may cause lung edema. Corrosive substances may cause lung damage (e.g. lung edema, fluid in the lungs). As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested. Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered. This must definitely be left to a doctor or person authorized by him/her. (ICSC13719).

## NOTES TO PHYSICIAN

for corrosives:

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BASIC TREATMENT

- Establish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary.
- Administer oxygen by non-rebreather mask at 10 t0 15 l/min.
- Monitor and treat, where necessary, for pulmonary edema .
- Monitor and treat, where necessary, for shock.
- Anticipate seizures .
- Where eyes have been exposed, flush immediately with water and continue to irrigate with normal saline during transport to hospital.
- DO NOT use emetics. Where ingestion is suspected rinse mouth and give up to 200 ml water (5 ml/kg recommended) for dilution where patient is able to swallow, has a strong gag reflex and does not drool.
- Skin burns should be covered with dry, sterile bandages, following decontamination.
- DO NOT attempt neutralization as exothermic reaction may occur.

ADVANCED TREATMENT

- Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
- Positive-pressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.

-----

- Start an IV D5W TKO. If signs of hypovolemia are present use lactated Ringers solution. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary edema.
- Hypotension with signs of hypovolemia requires the cautious administration of fluids. Fluid overload might create complications.
- Treat seizures with diazepam.
- Proparacaine hydrochloride should be used to assist eye irrigation.

EMERGENCY DEPARTMENT

- Laboratory analysis of complete blood count, serum electrolytes, BUN, creatinine, glucose, urinalysis, baseline for serum aminotransferases (ALT and AST), calcium, phosphorus and magnesium, may assist in establishing a treatment regime.
- Positive end-expiratory pressure (PEEP)-assisted ventilation may be required for acute parenchymal injury or adult respiratory distress syndrome.
- Consider endoscopy to evaluate oral injury.
- Consult a toxicologist as necessary.

BRONSTEIN, A.C. and CURRANCE, P.L. EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994. For exposures involving sulfides and hydrogen sulfide (including gastric acid decomposition products of alkaline sulfides).

- Hydrogen sulfide anion produces its major toxic effect through inhibition of cytochrome oxidases.
- Symptoms include profuse salivation, nausea, vomiting and diarrhea. Central nervous effects may include giddiness, headache, vertigo, amnesia, confusion and unconsciousness. Tachypnea, palpitation, tachycardia, arrhythmia, sweating,

weakness and muscle cramps may also indicate over-exposures.

Treatment involves:

- If respirations are depressed, application of artificial respiration, administration of oxygen (continue after spontaneous breathing is established).
- For severe poisonings administer amyl nitrite and sodium nitrite (as for cyanide poisoning) but omit sodium thiosulfate injection.
- Atropine sulfate (0.6 mg intramuscularly) may contribute symptomatic relief.
- Conjunctivitis may be relieved by installation of 1 drop of olive-oil in each eye and sometimes by 3 drops of epinephrine solution (1:1000) at frequent intervals. Occasionally local anesthetics and hot and cold compresses are necessary to control pain.
- Antibiotics at first hint of pulmonary infection.

[Gosselin et al, Clinical Toxicology of Commercial Products].

# Section 5 - FIRE FIGHTING MEASURES

Vapor Pressure (mmHg):	0.398 @ 25C
Upper Explosive Limit (%):	Not available
Specific Gravity (water=1):	1.32 @ 20C
Lower Explosive Limit (%):	5.9

#### **EXTINGUISHING MEDIA**

- - Water spray or fog.
- Foam.
- Dry chemical powder.
- BCF (where regulations permit).
- Carbon dioxide.

# FIRE FIGHTING

- •
- Alert Emergency Responders and tell them location and nature of hazard.
- Wear full body protective clothing with breathing apparatus.
- · Prevent, by any means available, spillage from entering drains or water course.
- Use fire fighting procedures suitable for surrounding area.
- · DO NOT approach containers suspected to be hot.
- · Cool fire exposed containers with water spray from a protected location.
- If safe to do so, remove containers from path of fire.
- Equipment should be thoroughly decontaminated after use.

## **GENERAL FIRE HAZARDS/HAZARDOUS COMBUSTIBLE PRODUCTS**

- · Combustible.
- Slight fire hazard when exposed to heat or flame.
- Heating may cause expansion or decomposition leading to violent rupture of containers.
- On combustion, may emit toxic fumes of carbon monoxide (CO).
- May emit acrid smoke.
- Mists containing combustible materials may be explosive.

Combustion products include: carbon dioxide (CO2), sulfur oxides (SOx), other pyrolysis products typical of burning organic material.

May emit corrosive fumes.

## FIRE INCOMPATIBILITY

Avoid contamination with oxidizing agents i.e. nitrates, oxidizing acids, chlorine bleaches, pool chlorine etc. as ignition may result.

## PERSONAL PROTECTION

Glasses: Chemical goggles. Full face- shield. Gloves: 1.BUTYL 2.NEOPRENE 3.VITON Respirator: Type A Filter of sufficient capacity

## Section 6 - ACCIDENTAL RELEASE MEASURES

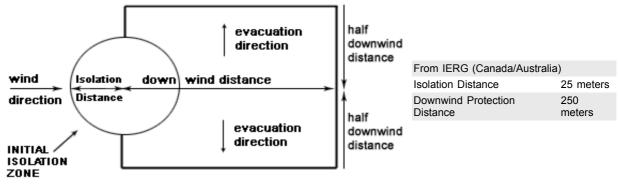
#### MINOR SPILLS

- •
- Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material.
- Check regularly for spills and leaks.
- Clean up all spills immediately.
- Avoid breathing vapors and contact with skin and eyes.
- · Control personal contact by using protective equipment.
- Contain and absorb spill with sand, earth, inert material or vermiculite.
- Wipe up.
- Place in a suitable labeled container for waste disposal.
- MAJOR SPILLS

- Clear area of personnel and move upwind.
- Alert Emergency Responders and tell them location and nature of hazard.
- Wear full body protective clothing with breathing apparatus.
- Prevent, by any means available, spillage from entering drains or water course.
- Consider evacuation.
- Stop leak if safe to do so.
- Contain spill with sand, earth or vermiculite.
- Collect recoverable product into labeled containers for recycling.
- Neutralize/decontaminate residue.
- Collect solid residues and seal in labeled drums for disposal.
- Wash area and prevent runoff into drains.
- After clean up operations, decontaminate and launder all protective clothing and equipment before storing and re-using.
- If contamination of drains or waterways occurs, advise emergency services.

## PROTECTIVE ACTIONS FOR SPILL

#### PROTECTIVE ACTION ZONE



#### FOOTNOTES

1 PROTECTIVE ACTION ZONE is defined as the area in which people are at risk of harmful exposure. This zone assumes that random changes in wind direction confines the vapour plume to an area within 30 degrees on either side of the predominant wind direction, resulting in a crosswind protective action distance equal to the downwind protective action distance. 2 PROTECTIVE ACTIONS should be initiated to the extent possible, beginning with those closest to the spill and working away from the site in the downwind

direction. Within the protective action zone a level of vapour concentration may exist resulting in nearly all unprotected persons becoming incapacitated and unable to take protective action and/or incurring serious or irreversible health effects.

INITIAL ISOLATION ZONE is determined as an area, including upwind of the incident, within which a high probability of localised wind reversal may expose nearly all persons without appropriate protection to life-threatening concentrations of the material.
 SMALL SPILLS involve a leaking package of 200 litres (55 US gallons) or less, such as a drum (jerrican or box with inner containers). Larger packages

leaking leaking leaking leaking so that a compressed gas leaking from a small cylinder are also considered "small spills". LARGE SPILLS involve many small leaking packages or a leaking package of greater than 200 litres, such as a cargo tank, portable tank or a "one-tonne" compressed gas cylinder.
5 Guide 153 is taken from the US DOT emergency response guide book.
6 IERG information is derived from CANUTEC - Transport Canada.

#### ACUTE EXPOSURE GUIDELINE LEVELS (AEGL) (in ppm)

AEGL 1: The airborne concentration of a substance above which it is predicted

that the general population, including susceptible individuals, could

experience notable discomfort, irritation, or certain asymptomatic nonsensory

effects. However, the effects are not disabling and are transient and

reversible upon cessation of exposure.

AEGL 2: The airborne concentration of a substance above which it is predicted

that the general population, including susceptible individuals, could

experience irreversible or other serious, long-lasting adverse health effects

or an impaired ability to escape.

AEGL 3: The airborne concentration of a substance above which it is predicted that the general population, including susceptible individuals, could experience life-threatening health effects or death.

# Section 7 - HANDLING AND STORAGE

## PROCEDURE FOR HANDLING

- DO NOT allow clothing wet with material to stay in contact with skin
- Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- WARNING: To avoid violent reaction, ALWAYS add material to water and NEVER water to material.
- Avoid smoking, naked lights or ignition sources. .
- Avoid contact with incompatible materials.
- When handling, DO NOT eat, drink or smoke.
- Keep containers securely sealed when not in use.
- Avoid physical damage to containers.
- Always wash hands with soap and water after handling.
- Work clothes should be laundered separately.
- Launder contaminated clothing before re-use.
- Use good occupational work practice.

- Observe manufacturer's storing and handling recommendations.
- Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.

# **RECOMMENDED STORAGE METHODS**

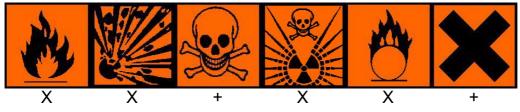
- Glass container.
- Lined metal can, Lined metal pail/drum
- Plastic pail
- Polyliner drum
- Packing as recommended by manufacturer.
- · Check all containers are clearly labeled and free from leaks.
- For low viscosity materials
- Drums and jerricans must be of the non-removable head type.
- Where a can is to be used as an inner package, the can must have a screwed enclosure.
- For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):
- · Removable head packaging;
- Cans with friction closures and
- low pressure tubes and cartridges may be used.

- Where combination packages are used, and the inner packages are of glass, porcelain or stoneware, there must be sufficient inert cushioning material in contact with inner and outer packages unless the outer packaging is a close fitting molded plastic box and the substances are not incompatible with the plastic.

## STORAGE REQUIREMENTS

- - Store in original containers.
- • Keep containers securely sealed.
- Store in a cool, dry, well-ventilated area. •
- Store away from incompatible materials and foodstuff containers.
- Protect containers against physical damage and check regularly for leaks. ٠
- ٠ Observe manufacturer's storing and handling recommendations.

# SAFE STORAGE WITH OTHER CLASSIFIED CHEMICALS



X: Must not be stored together

O: May be stored together with specific preventions

+: May be stored together

# Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

## **EXPOSURE CONTROLS**

Source	Material	TWA ppm	TWA mg/m³		Peak mg/m³	Notes
Canada - British Columbia Occupational Exposure Limits	thioglycolic acid (Thioglycolic acid)	1				Skin
Canada - Ontario Occupational Exposure Limits	thioglycolic acid (Thioglycolic acid)	1	3.8			Skin
Canada - Alberta Occupational Exposure Limits	thioglycolic acid (Thioglycolic acid)	1	3.8			
US ACGIH Threshold Limit Values (TLV)	thioglycolic acid (Thioglycolic acid)	1				TLV Basis: eye & skin irritation
US NIOSH Recommended Exposure Limits (RELs)	thioglycolic acid (Thioglycolic acid)	1	4			
US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants	thioglycolic acid (Thioglycolic acid)	1	4			
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	thioglycolic acid (Thioglycolic acid)	1	4			
US - Minnesota Permissible Exposure Limits (PELs)	thioglycolic acid (Thioglycolic acid)	1	4			

US - California Permissible Exposure Limits for Chemical Contaminants	thiogrycolic acid (Thioglycolic acid)	1	3.8			
US - Hawaii Air Contaminant Limits	thioglycolic acid (Thioglycolic acid)	1	4			
US - Alaska Limits for Air Contaminants	thioglycolic acid (Thioglycolic acid)	1	4			
Canada - Saskatchewan Occupational Health and Safety Regulations - Contamination Limits	thioglycolic acid (Thioglycolic acid)	1		2		Skin
Canada - Yukon Permissible Concentrations for Airborne Contaminant Substances	thioglycolic acid (Thioglycolic acid)	1	5	2	10	
US - Washington Permissible exposure limits of air contaminants	thioglycolic acid (Thioglycolic acid)	1		3		
Canada - Prince Edward Island Occupational Exposure Limits	thioglycolic acid (Thioglycolic acid)	1				TLV Basis: eye & skin irritation
Canada - Quebec Permissible Exposure Values for Airborne Contaminants (English)	thioglycolic acid (Thioglycolic acid)	1	3.8			
Canada - Nova Scotia Occupational Exposure Limits	thioglycolic acid (Thioglycolic acid)	1				TLV Basis: eye & skin irritation
						initation
US - Michigan Exposure Limits for Air Contaminants	thioglycolic acid (Thioglycolic acid)	1	4			intation

## **MATERIAL DATA**

THIOGLYCOLIC ACID:

■ For thioglycolic acid: Exposure at or below the TLV-TWA is thought to protect the worker from the significant risk of eye and skin irritation and systemic effects. The skin notation recognises that percutaneous absorption has caused systemic toxicity in experimental animals.

## PERSONAL PROTECTION



Consult your EHS staff for recommendations

#### EYE

- .
- Chemical goggles.
- Full face shield.
- Contact lenses pose a special hazard; soft contact lenses may absorb irritants and all lenses concentrate them.

# HANDS/FEET

Elbow length PVC gloves.

• When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.

NOTE: The material may produce skin sensitization in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include: such as: • frequency and duration of contact,

- chemical resistance of glove material,
- glove thickness and
- dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739).

- When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374) is recommended.
- When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374) is recommended.
- Contaminated gloves should be replaced.

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a

non-perfumed moisturiser is recommended.

#### OTHER

- Overalls.
- ٠ PVC Apron.
- PVC protective suit may be required if exposure severe.
- Evewash unit.
- Ensure there is ready access to a safety shower.

#### **GLOVE SELECTION INDEX**

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the computer-generated selection: thioglycolic acid

Protective Material CPI *.	
BUTYL	A
NEOPRENE	A
VITON	A
NITRILE	С
* CPI - Chemwatch Performance Index	

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation.

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

#### RESPIRATOR

Selection of the Class and Type of respirator will depend upon the level of breathing zone contaminant and the chemical nature of the contaminant. Protection Factors (defined as the ratio of contaminant outside and inside the mask) may also be important.

Breathing Zone Level ppm (volume)	Maximum Protection Factor	Half-face Respirator	Full-Face Respirator
1000	10	A-1	-
1000	50	-	A-1
5000	50	Airline*	-
5000	100	-	A-2
10000	100	-	A-3
	100+		Airline* *

\* - Continuous Flow \*\* - Continuous-flow or positive pressure demand.

The local concentration of material, quantity and conditions of use determine the type of personal protective equipment required.

Use appropriate NIOSH-certified respirator based on informed professional judgement. In conditions where no reasonable estimate of exposure can be made, assume the exposure is in a concentration IDLH and use NIOSH-certified full face pressure demand SCBA with a minimum service life of 30 minutes, or a combination full facepiece pressure demand SAR with auxiliary self-contained air supply. Respirators provided only for escape from IDLH atmospheres shall be NIOSH-certified for escape from the atmosphere in which they will be used.

#### **ENGINEERING CONTROLS**

Lower end of the range

1: Room air currents minimal or favorable to capture

2: Contaminants of low toxicity or of nuisance value only.

Local exhaust ventilation usually required. If risk of overexposure exists, wear an approved respirator. Correct fit is essential to obtain adequate protection an approved self contained breathing apparatus (SCBA) may be required in some situations. Provide adequate ventilation in warehouse or closed storage area.

"escape" velocities which, in turn, determine the "capture contaminants generated in the workplace possess varving

Upper end of the range

1: Disturbing room air currents

2: Contaminants of high toxicity

velocities" of fresh circulating air required to effectively remove	
Type of Contaminant:	Air Speed:
solvent, vapors, degreasing etc., evaporating from tank (in still air).	0.25-0.5 m/s (50-100 f/min.)
aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)	0.5-1 m/s (100-200 f/min.)
direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)
grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).	2.5-10 m/s (500-2000 f/min.)
Within each range the appropriate value depends on:	

#### 3: Intermittent, low production.

4: Large hood or large air mass in motion

#### 3: High production, heavy use

4: Small hood-local control only

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

# **Section 9 - PHYSICAL AND CHEMICAL PROPERTIES**

## PHYSICAL PROPERTIES

Liquid. Mixes with water. Corrosive. Toxic or noxious vapors/ gas. Contact with acids liberates very toxic gas. Molecular Weight 92 11 State Liquid Melting Range (°F) 23 Viscosity Not Available Boiling Range (°F) 253.4 Solubility in water (g/L) Miscible Flash Point (°F) >233.6 pH (1% solution) Not available. Decomposition Temp (°F) Not Available pH (as supplied) Not available Autoignition Temp (°F) 662 Vapor Pressure (mmHg) 0.398 @ 25C Upper Explosive Limit (%) Not available Specific Gravity (water=1) 1.32 @ 20C Lower Explosive Limit (%) 59 Relative Vapor Density (air=1) 32 Volatile Component (%vol) 100 **Evaporation Rate** Not available

#### APPEARANCE

Colourless liquid with a strong, very unpleasant odour. Miscible with water, alcohol, ether, chloroform, benzene and many other organic solvents. Corrosive to metals.

## Section 10 - CHEMICAL STABILITY

## CONDITIONS CONTRIBUTING TO INSTABILITY

- Presence of incompatible materials.
- Product is considered stable.
- · Hazardous polymerization will not occur.

#### STORAGE INCOMPATIBILITY

Avoid strong acids.

Avoid reaction with oxidizing agents.

For incompatible materials - refer to Section 7 - Handling and Storage.

## Section 11 - TOXICOLOGICAL INFORMATION

thioglycolic acid

## TOXICITY AND IRRITATION

unless otherwise specified data extracted from RTECS - Register of Toxic Effects of Chemical Substances.
 TOXICITY
 IRRITATION

Oral (rat) LD50: 114 mg/kg

Skin (human): 3%

Oral (mouse) LD50: 242 mg/kg

Dermal (rabbit) LDLo: 300 mg/kg

■ Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. RADS (or asthma) following an irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production.

Ammonium and glyceryl thioglycolate and thioglycolic acid are used predominantly in cosmetic permanent waving lotions at concentrations up to 15.4% (as thioglycolic ccid). At use concentrations, these cosmetic ingredients are only slightly toxic in acute single oral and dermal exposures. In repeated dermal tests for extended periods of exposure, these ingredients were toxic. Commercial permanent wave products produced transient conjunctival redness to both rinsed and unrinsed eyes.

The results of skin testing for irritation and sensitisation of these thioglycolates depends on the type of test system used. Under

occlusive patch testing, the data indicate that these ingredients are cumulative irritants and possibly weak sensitisers, but not under semi-occlusive test conditions. In clinical patients, mainly hairdressers, glyceryl thioglycolate elicited allergic reactions at concentrations down to 0.25%.

Administration of ammonium thioglycolate to animals has produced hypoglycaemia, and thyroid effects.

The acute inhalation toxicity of a liquid droplet aerosol containing aqueous ammonium thioglycolate (as 60% thioglycolic acid) was evaluated using rats (number and strain not stated). Animals were exposed to the aerosol for 1 hr and then observed for 14 days. None of the animals died. Few animals experienced respiratory distress, and signs were not observed beyond 24 hr post-exposure. At necropsy, minor pulmonary abnormalities were observed

Hair-waving solutions containing thioglycolates may produce keratitis; they may also produce irritation, burning sensations, conjunctival inflammation, corneal epithelial erosion, turbidity of the cornea, mydriasis, cytoplegia, loss of convergence and disturbances of vision. A mother and daughter developed bilateral optic neuritis after the use of a cold wave lotion containing ammonia thioglycolate . Oedema of the discs and retina and centrocecal scotomas horizontal oval defect in the field of vision) were seen. Oedema subsided in six months but the scotomas persisted.

Sodium thioglycolate, which has widespread occupational and consumer exposure to women from cosmetics and hair-care products, was evaluated for developmental toxicity by topical exposure during the embryonic and fetal periods of pregnancy . In rats, maternal topical exposure to sodium thioglycolate, at 200 mg/kg/day (the highest dose tested) on gestational days (GD) 6-19, resulted in maternal toxicity, including reduced body weights and weight gain, increased relative water consumption and one death. Treatment-related increases in feed consumption and changes at the application site occurred at all doses, in the absence of increased body weights or body weight change. Foetal body weights/litter were decreased at 200 mg/kg/day, with no other embryo/foetal toxicity and no treatment-related teratogenicity in any group. In rabbits, maternal topical exposure to sodium thioglycolate on GD 6-29 resulted in maternal dose-related toxicity at the dosing site in all groups; no maternal systemic toxicity, embryo/fetal toxicity, or treatment-related teratogenicity were observed in any group. A no observed adverse effect level (NOAEL) was not identified for maternal toxicity in either species with the dosages tested. The developmental toxicity NOAEL was 100 mg/kg/day (rats) and >65 mg/kg/day (rabbits; the highest dose tested). The clinical relevance of theses study results is uncertain because no data were available for levels, frequency, or duration of exposures in female workers or end users.

The material may produce severe irritation to the eye causing pronounced inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis.

The material may produce respiratory tract irritation, and result in damage to the lung including reduced lung function.

The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.

SKIN

SKIN			
thioglycolic acid	Canada - Ontario Occupational Exposure Limits - Skin	Notes	Skin
thioglycolic acid	US AIHA Workplace Environmental Exposure Levels (WEELs) - Skin	Notes	Skin
thioglycolic acid	US NIOSH Recommended Exposure Limits (RELs) - Skin	Skin	Yes
thioglycolic acid	Canada - Quebec Permissible Exposure Values for Airborne Contaminants - Skin (French)	Notes	Skin
thioglycolic acid	US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants - Skin	Skin Designation	х
thioglycolic acid	US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants - Skin	Skin Designation	х
thioglycolic acid	US - Washington Permissible exposure limits of air contaminants - Skin	Skin	х
thioglycolic acid	US ACGIH Threshold Limit Values (TLV) - Skin	Skin Designation	Yes
thioglycolic acid	Canada - British Columbia Occupational Exposure Limits - Skin	Notation	Skin
thioglycolic acid	US - Minnesota Permissible Exposure Limits (PELs) - Skin	Skin Designation	х
thioglycolic acid	US - Hawaii Air Contaminant Limits - Skin Designation	Skin Designation	х
thioglycolic acid	ND	Skin Designation	х
thioglycolic acid	US OSHA Permissible Exposure Levels (PELs) - Skin	Skin Designation	х
thioglycolic acid	US - California Permissible Exposure Limits for Chemical Contaminants - Skin	Skin	х
thioglycolic acid	US - California Permissible Exposure Limits for Chemical Contaminants - Skin	Skin	S
thioglycolic acid	Canada - Alberta Occupational Exposure Limits - Skin	Substance Interaction	1,3

# Section 12 - ECOLOGICAL INFORMATION

Refer to data for ingredients, which follows:

THIOGLYCOLIC AČID:

Prevent, by any means available, spillage from entering drains or watercourses.
 DO NOT discharge into sewer or waterways.

## Ecotoxicity

Ingredient thioglycolic acid Persistence: Water/Soil Persistence: Air LOW

Bioaccumulation LOW

Mobility HIGH

## Section 13 - DISPOSAL CONSIDERATIONS

## **US EPA Waste Number & Descriptions**

## A. General Product Information

Corrosivity characteristic: use EPA hazardous waste number D002 (waste code C)

## Reactivity characteristic: use EPA hazardous waste number D003 (waste code R).

#### **Disposal Instructions**

All waste must be handled in accordance with local, state and federal regulations.

Puncture containers to prevent re-use and bury at an authorized landfill.

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- Reduction,
- Reuse
- Recycling
- Disposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.

DO NOT allow wash water from cleaning equipment to enter drains. Collect all wash water for treatment before disposal.

- Recycle wherever possible.
- Consult manufacturer for recycling options or consult Waste Management Authority for disposal if no suitable treatment or disposal facility can be identified.
- Treat and neutralize at an approved treatment plant.
- Treatment should involve: Neutralization followed by: Burial in a licensed land-fill or Incineration in a licensed apparatus
- Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.

# **Section 14 - TRANSPORTATION INFORMATION**



DOT:							
Symbols:	None	Hazard class or Division:	8				
Identification Numbers:	UN1940	PG:	II				
Label Codes:	8	Special provisions:	A7, B2, IB2, N34, T7, TP2				
Packaging: Exceptions:	154	Packaging: Non-bulk:	202				
Packaging: Exceptions:	154	Quantity limitations: Passenger aircraft/rail:	1 L				
Quantity Limitations: Cargo aircraft only:	30 L	Vessel stowage: Location:	А				
Vessel stowage: Other:	None						
Hazardous materials descriptions and proper shipping names: Thioglycolic acid <b>Air Transport IATA:</b>							
ICAO/IATA Class:	8	ICAO/IATA Subrisk:	None				
UN/ID Number:	1940	Packing Group:	II				
Special provisions:	None						
Shipping Name: THIOGLYCOLIC ACID Maritime Transport IMDG:							
IMDG Class:	8	IMDG Subrisk:	None				
UN Number:	1940	Packing Group:	II				
EMS Number:	F-A,S-B	Special provisions:	None				
Limited Quantities: Shipping Name: THIOGLYCOL							

## Section 15 - REGULATORY INFORMATION

#### thioglycolic acid (CAS: 68-11-1) is found on the following regulatory lists;

"Canada - Alberta Occupational Exposure Limits", "Canada - British Columbia Occupational Exposure Limits", "Canada - Northwest Territories Occupational Exposure Limits (English)", "Canada - Nova Scotia Occupational Exposure Limits", "Canada - Ontario Occupational Exposure Limits", "Canada - Prince Edward Island Occupational Exposure Limits", "Canada - Quebec Permissible Exposure Values for Airborne Contaminants (English)", "Canada - Saskatchewan Occupational Health and Safety

Regulations - Contamination Limits", "Canada - Yukon Permissible Concentrations for Airborne Contaminant Substances", "Canada Ingredient Disclosure List (SOR/88-64)", "Canada Toxicological Index Service - Workplace Hazardous Materials Information System - WHMIS (English)", "Canada Toxicological Index Service - Workplace Hazardous Materials Information System - WHMIS (French)", "International Council of Chemical Associations (ICCA) - High Production Volume List", "OECD Representative List of High Production Volume (HPV) Chemicals", "US - Alaska Limits for Air Contaminants", "US - California Occupational Safety and Health Regulations (CAL/OSHA) - Hazardous Substances List", "US - California Permissible Exposure Limits for Chemical Contaminants", "US - Connecticut Hazardous Air Pollutants", "US - Hawaii Air Contaminants", "US - Massachusetts Oil & Hazardous Material List", "US - Michigan Exposure Limits for Air Contaminants", "US - Minnesota Hazardous Substance List", "US - New Jersey Right to Know Hazardous Substances", "US - Pennsylvania - Hazardous Substance List", "US - New Jersey Right to Know Hazardous Substances", "US - Pennsylvania - Hazardous Substance List", "US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants", "US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants", "US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants", "US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants", "US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits (IRC OPERCIPATION Permissible exposure Limits of air contaminants", "US ACGIH Threshold Limit Values (TLV)", "US Cosmetic Ingredient Review (CIR) Cosmetic ingredients found safe, with qualifications", "US NIOSH Recommended Exposure Limits (RELs)", "US FOA High Production Volume Program Chemical List", "US NIOSH Recommended Exposure Limits (RELs)", "US Toxic Substances Control Act (TSCA) - Inventory"

## **Section 16 - OTHER INFORMATION**

#### LIMITED EVIDENCE

Cumulative effects may result following exposure\*.

Possible skin sensitizer\*.

\* (limited evidence).

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■ Classification of the mixture and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references. A list of reference resources used to assist the committee may be found at: www.chemwatch.net/references.

■ The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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