# Gadolinium (III) chloride hexahydrate

sc-202617

**Material Safety Data Sheet** 



The Power to Oscotion

Hazard Alert Code Key:

**EXTREME** 

HIGH

MODERATE

LOW

# Section 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

# **PRODUCT NAME**

Gadolinium (III) chloride hexahydrate

# STATEMENT OF HAZARDOUS NATURE

CONSIDERED A HAZARDOUS SUBSTANCE ACCORDING TO OSHA 29 CFR 1910.1200.

# **NFPA**



# **SUPPLIER**

Santa Cruz Biotechnology, Inc. 2145 Delaware Avenue Santa Cruz, California 95060 800.457.3801 or 831.457.3800

## EMERGENCY:

ChemWatch

Within the US & Canada: 877-715-9305 Outside the US & Canada: +800 2436 2255 (1-800-CHEMCALL) or call +613 9573 3112

# **SYNONYMS**

GdCl3 • 6H2O, "gadolinium trichloride"

# **Section 2 - HAZARDS IDENTIFICATION**

## **CHEMWATCH HAZARD RATINGS**

| OHEMINATOH HAZARD RATINGO |   |     |                     |  |  |  |  |
|---------------------------|---|-----|---------------------|--|--|--|--|
|                           |   | Min | Max                 |  |  |  |  |
| Flammability:             | 0 |     |                     |  |  |  |  |
| Toxicity:                 | 2 |     |                     |  |  |  |  |
| Body Contact:             | 2 |     | Min/Nil=0<br>Low=1  |  |  |  |  |
| Reactivity:               | 0 |     | Moderate=2          |  |  |  |  |
| Chronic:                  | 2 |     | High=3<br>Extreme=4 |  |  |  |  |

#### **CANADIAN WHMIS SYMBOLS**



#### **RISK**

Irritating to eyes, respiratory system and skin.

### **POTENTIAL HEALTH EFFECTS**

### **ACUTE HEALTH EFFECTS**

#### **SWALLOWED**

- Accidental ingestion of the material may be damaging to the health of the individual.
- Lanthanide poisoning causes immediate defecation, writhing, incoordination, labored breathing, and inactivity. Respiratory and heart failure may follow causing death.

#### **EYE**

■ Evidence exists, or practical experience predicts, that the material may cause eye irritation in a substantial number of individuals. Prolonged eye contact may cause inflammation characterized by a temporary redness of the conjunctiva (similar to windburn).

#### SKIN

- The material may cause moderate inflammation of the skin either following direct contact or after a delay of some time. Repeated exposure can cause contact dermatitis which is characterized by redness, swelling and blistering.
- Skin contact is not thought to have harmful health effects, however the material may still produce health damage following entry through wounds, lesions or abrasions.
- Open cuts, abraded or irritated skin should not be exposed to this material.
- Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

### **INHALED**

- The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.
- Persons with impaired respiratory function, airway diseases and conditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled.
- Exposure to vapors of some rare earth salts can cause sensitivity to heat, itching, and increased sensitivity of smell and taste. Other effects include inflamed airways and lung, emphysema, regional narrowing of terminal airways and cell changes.

#### **CHRONIC HEALTH EFFECTS**

■ Long-term exposure to respiratory irritants may result in disease of the airways involving difficult breathing and related systemic problems. Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems.

Long term exposure to high dust concentrations may cause changes in lung function i.e. pneumoconiosis; caused by particles less than 0.5 micron penetrating and remaining in the lung.

Gadolinium is a rare earth metal - heavy type (yttrium family). There has been no reports of poisoning in workers, although the metal can cause chest X-ray abnormalities due to its high density.

# Section 3 - COMPOSITION / INFORMATION ON INGREDIENTS

| NAME                     | CAS RN     | %    |
|--------------------------|------------|------|
| gadolinium(III) chloride | 10138-52-0 | > 99 |

## **Section 4 - FIRST AID MEASURES**

# **SWALLOWED**

· If swallowed do NOT induce vomiting. · If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.

## EYE

■ If this product comes in contact with the eyes: · Wash out immediately with fresh running water. · Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.

#### SKIN

■ If skin contact occurs: · Immediately remove all contaminated clothing, including footwear · Flush skin and hair with running water (and soap if available).

## **INHALED**

· If fumes or combustion products are inhaled remove from contaminated area. · Lay patient down. Keep warm and rested.

# **NOTES TO PHYSICIAN**

■ Treat symptomatically.

| Section 5 - FIRE FIGHTING MEASURES |                  |  |  |  |  |
|------------------------------------|------------------|--|--|--|--|
| Vapour Pressure (mmHG):            | Not applicable.  |  |  |  |  |
| Upper Explosive Limit (%):         | Not applicable   |  |  |  |  |
| Specific Gravity (water=1):        | 4.52; 2.42(6H2O) |  |  |  |  |
| Lower Explosive Limit (%):         | Not applicable   |  |  |  |  |

# **EXTINGUISHING MEDIA**

· There is no restriction on the type of extinguisher which may be used.

Use extinguishing media suitable for surrounding area.

### **FIRE FIGHTING**

- · Alert Emergency Responders and tell them location and nature of hazard.
- · Wear breathing apparatus plus protective gloves for fire only.

# GENERAL FIRE HAZARDS/HAZARDOUS COMBUSTIBLE PRODUCTS

- · Non combustible.
- · Not considered to be a significant fire risk, however containers may burn.

Decomposition may produce toxic fumes of: hydrogen chloride, metal oxides.

May emit poisonous fumes.

May emit corrosive fumes.

## FIRE INCOMPATIBILITY

None known.

# PERSONAL PROTECTION

Glasses:

Safety Glasses.

Chemical goggles.

Gloves:

Respirator:

Particulate

# Section 6 - ACCIDENTAL RELEASE MEASURES

#### MINOR SPILLS

- · Remove all ignition sources.
- · Clean up all spills immediately.
- · Avoid contact with skin and eyes.
- · Control personal contact by using protective equipment.
- · Use dry clean up procedures and avoid generating dust.
- · Place in a suitable, labelled container for waste disposal.

#### **MAJOR SPILLS**

- Moderate hazard.
- · CAUTION: Advise personnel in area.
- · Alert Emergency Responders and tell them location and nature of hazard.

# **Section 7 - HANDLING AND STORAGE**

# PROCEDURE FOR HANDLING

- · Avoid all personal contact, including inhalation.
- · Wear protective clothing when risk of exposure occurs.

## RECOMMENDED STORAGE METHODS

- Glass container.
- · Polyethylene or polypropylene container.
- Check all containers are clearly labelled and free from leaks.

# STORAGE REQUIREMENTS

- · Store in original containers.
- · Keep containers securely sealed.

# Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

### **EXPOSURE CONTROLS**

The following materials had no OELs on our records

• gadolinium(III) chloride: CAS:10138-52-0 CAS:13450-84-5

## PERSONAL PROTECTION



## **RESPIRATOR**

Particulate

Consult your EHS staff for recommendations

#### **EYE**

- · Safety glasses with side shields.
- · Chemical goggles.

#### HANDS/FEET

- Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include: such as:
- · frequency and duration of contact,
- · chemical resistance of glove material,
- · glove thickness and
- · dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739).

- · When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374) is recommended.
- · When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374) is recommended.
- · Contaminated gloves should be replaced.

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

Experience indicates that the following polymers are suitable as glove materials for protection against undissolved, dry solids, where abrasive particles are not present.

- · polychloroprene
- · nitrile rubber
- · butyl rubber
- fluorocaoutchouc
- · polyvinyl chloride

Gloves should be examined for wear and/ or degradation constantly.

#### **OTHER**

- · Overalls.
- · P.V.C. apron.
- · Barrier cream.
- · Skin cleansing cream.
- · Eye wash unit.

#### **ENGINEERING CONTROLS**

- · Local exhaust ventilation is required where solids are handled as powders or crystals; even when particulates are relatively large, a certain proportion will be powdered by mutual friction.
- · If in spite of local exhaust an adverse concentration of the substance in air could occur, respiratory protection should be considered.

# **Section 9 - PHYSICAL AND CHEMICAL PROPERTIES**

### PHYSICAL PROPERTIES

Solid.

Mixes with water.

| State                     | Divided solid   | Molecular Weight               | 371.7 (.6H2O)    |
|---------------------------|-----------------|--------------------------------|------------------|
| Melting Range (°F)        | 1128.2~         | Viscosity                      | Not Applicable   |
| Boiling Range (°F)        | Not applicable. | Solubility in water (g/L)      | Miscible         |
| Flash Point (°F)          | Not applicable  | pH (1% solution)               | < 7              |
| Decomposition Temp (°F)   | Not available   | pH (as supplied)               | Not applicable   |
| Autoignition Temp (°F)    | Not applicable  | Vapour Pressure (mmHG)         | Not applicable.  |
| Upper Explosive Limit (%) | Not applicable  | Specific Gravity (water=1)     | 4.52; 2.42(6H2O) |
| Lower Explosive Limit (%) | Not applicable  | Relative Vapor Density (air=1) | Not applicable.  |
| Volatile Component (%vol) | Not applicable. | Evaporation Rate               | Not applicable   |

# **APPEARANCE**

White crystalline hygroscopic powder; mixes with water.

# **Section 10 - CHEMICAL STABILITY**

# **CONDITIONS CONTRIBUTING TO INSTABILITY**

- · Presence of incompatible materials.
- · Product is considered stable.

## STORAGE INCOMPATIBILITY

- · Metals and their oxides or salts may react violently with chlorine trifluoride and bromine trifluoride.
- These trifluorides are hypergolic oxidisers. They ignites on contact (without external source of heat or ignition) with recognised fuels contact with these materials, following an ambient or slightly elevated temperature, is often violent and may produce ignition.
- · The state of subdivision may affect the results.
- · Avoid strong bases.

# Section 11 - TOXICOLOGICAL INFORMATION

GADOLINIUM(III) CHLORIDE

### **TOXICITY AND IRRITATION**

GADOLINIUM(III) CHLORIDE:

■ unless otherwise specified data extracted from RTECS - Register of Toxic Effects of Chemical Substances.

TOXICITY IRRITATION

Intraperitoneal (mouse) LD50: 378 mg/kg Skin (rabbit): 500 mg - Moderate

Intraperitoneal (mouse) LD50: 669 mg/kg Eye (rabbit): 1 mg/h - Mild

Intravenous (mouse) LD50: 100 mg/kg Nil Reported

■ Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production.

Lanthanide poisoning causes immediate defecation, writhing, incoordination, labored breathing, and inactivity. Respiratory and heart failure may follow causing death.

The material may be irritating to the eye, with prolonged contact causing inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis.

The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.

#### anhydrous:

Somnolence, respiratory changes, food intake changes recorded.

hydrate:

## Section 12 - ECOLOGICAL INFORMATION

No data

**Ecotoxicity** 

Ingredient Persistence: Water/Soil Persistence: Air Bioaccumulation Mobility gadolinium(III) chloride HIGH LOW HIGH

#### Section 13 - DISPOSAL CONSIDERATIONS

# **Disposal Instructions**

All waste must be handled in accordance with local, state and federal regulations.

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- · Reduction
- · Reuse
- · Recycling
- · Disposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.

DO NOT allow wash water from cleaning equipment to enter drains. Collect all wash water for treatment before disposal. For small quantities:

- · Carefully make a 5% of the solution in water or dilute acid controllingany vigorous exotherm or fumes by rate of addition and cooling.
- $\cdot$  Recycle wherever possible or consult manufacturer for recycling options.
- · Consult Waste Management Authority for disposal.

# **Section 14 - TRANSPORTATION INFORMATION**

NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS: DOT, IATA, IMDG

# Section 15 - REGULATORY INFORMATION

## gadolinium(III) chloride (CAS: 10138-52-0,13450-84-5) is found on the following regulatory lists;

"Canada Domestic Substances List (DSL)", "Canada Ingredient Disclosure List (SOR/88-64)", "US Toxic Substances Control Act (TSCA) - Inventory"

### **Section 16 - OTHER INFORMATION**

## Ingredients with multiple CAS Nos

Ingredient Name CAS gadolinium(III) chloride 10138-52-0, 13450-84-5

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- Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

  A list of reference resources used to assist the committee may be found at:

  www.chemwatch.net/references.
- The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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