# Epibromohydrin



# Section 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

# PRODUCT NAME

#### Epibromohydrin

## STATEMENT OF HAZARDOUS NATURE

CONSIDERED A HAZARDOUS SUBSTANCE ACCORDING TO OSHA 29 CFR 1910.1200.



# SUPPLIER

Santa Cruz Biotechnology, Inc. 2145 Delaware Avenue Santa Cruz, California 95060 800.457.3801 or 831.457.3800 **EMERGENCY** ChemWatch Within the US & Canada: 877-715-9305 Outside the US & Canada: +800 2436 2255 (1-800-CHEMCALL) or call +613 9573 3112

## SYNONYMS

C3-H5-Br-O, "propane, 3-bromo-1, 2-epoxy-", "3-bromo-1, 2-epoxypropane", "1-bromo-2, 3-epoxypropane", epibromhydrin, epibromhydrine

#### **CHEMWATCH HAZARD RATINGS** Min Max Flammability: 4 Toxicity: 4 Min/Nil=0 Body Contact: 4 Low=1 Moderate=2 Reactivity: 1 High=3 Chronic: 3 Extreme=4

# CANADIAN WHMIS SYMBOLS



EMERGENCY OVERVIEW

# Section 2 - HAZARDS IDENTIFICATION

## RISK

Causes burns. Risk of serious damage to eyes. Limited evidence of a carcinogenic effect. Very toxic by inhalation, in contact with skin and if swallowed. Flammable. Harmful to aquatic organisms.

#### POTENTIAL HEALTH EFFECTS

#### ACUTE HEALTH EFFECTS

#### SWALLOWED

• Severely toxic effects may result from the accidental ingestion of the material; animal experiments indicate that ingestion of less than 5 gram may be fatal or may produce serious damage to the health of the individual.

• The material can produce chemical burns within the oral cavity and gastrointestinal tract following ingestion.

#### EYE

The material can produce chemical burns to the eye following direct contact.

- Vapors or mists may be extremely irritating.
- If applied to the eyes, this material causes severe eye damage.
- Irritation of the eyes may produce a heavy secretion of tears (lachrymation).

#### SKIN

Skin contact with the material may produce severely toxic effects; systemic effects may result following absorption and these may be fatal.

- The material can produce chemical burns following direct contactwith the skin.
- Open cuts, abraded or irritated skin should not be exposed to this material.
- Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects.
- Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

This material can cause inflammation of the skin oncontact in some persons.

#### INHALED

Inhalation of vapors or aerosols (mists, fumes), generated by the material during the course of normal handling, may produce severely toxic effects; these may be fatal.

Inhalation of vapours may cause drowsiness and dizziness.

- This may be accompanied by narcosis, reduced alertness, loss of reflexes, lack of coordination and vertigo.
- The material can cause respiratory irritation in some persons.
- The body's response to such irritation can cause further lung damage.

A single human exposure to epichlorohydrin is said to have resulted in chronic bronchitis and severe fatty infiltration of the liver.

Inhalation may result in nose and throat irritation, headache, nausea, vomiting and central nervous system depression.

#### **CHRONIC HEALTH EFFECTS**

There has been concern that this material can cause cancer or mutations, but there is not enough data to make an assessment.

Repeated or prolonged exposure to corrosives may result in the erosion of teeth, inflammatory and ulcerative changes in the mouth and necrosis (rarely) of the jaw. Bronchial irritation, with cough, and frequent attacks of bronchial pneumonia may ensue.

Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems.

There is some evidence that human exposure to the material may result in developmental toxicity. This evidence is based on animal studies where effects have been observed in the absence of marked maternal toxicity, or at around the same dose levels as other toxic effects but which are not secondary non-specific consequences of the other toxic effects.

Chronic intoxication with ionic bromides, historically, has resulted from medical use of bromides but not from environmental or occupational exposure; depression, hallucinosis, and schizophreniform psychosis can be seen in the absence of other signs of intoxication. Bromides may also induce sedation, irritability, agitation, delirium, memory loss, confusion, disorientation, forgetfulness (aphasias), dysarthria, weakness, fatigue, vertigo, stupor, coma, decreased appetite, nausea and vomiting, diarrhoea, hallucinations, an acne like rash on the face, legs and trunk, known as bronchoderma (seen in 25-30% of case involving bromide ion), and a profuse discharge from the nostrils (coryza). Ataxia and generalised hyperreflexia have also been observed. Correlation of neurologic symptoms with blood levels of bromide is inexact. The use of substances such as brompheniramine, as antihistamines, largely reflect current day usage of bromides; ionic bromides have been largely withdrawn from therapeutic use due to their toxicity. Several cases of foetal abnormalities have been described in mothers who took large doses of bromides during pregnancy.

When administered by gavage, epichlorohydrin induced carcinomas and papillomas of the fore-stomach in male rats. When administered by inhalation, the compound induced carcinomas and papillomas of the nasal cavity in male rats. Subcutaneous injection produced local sarcomas in female mice. Epichlorohydrin also was active as an initiator in the two-stage skin carcinogenesis study of female mice but did not induce tumors when applied alone to the skin. One study of factory workers shows a significant excess of respiratory cancer. However, some of the workers may also have been exposed to diisopropyl sulfate.

The epoxide group is an alkylating agent and thus destroysnucleotides within the cell. This may cause cancer.

Section 3 - COMPOSITION / INFORMATION ON INGREDIENTS					
NAME	CAS	S RN	%		
epibromohydrin	3132	2-64-7	>98		

# Section 4 - FIRST AID MEASURES

#### SWALLOWED

· For advice, contact a Poisons Information Center or a doctor at once. · Urgent hospital treatment is likely to be needed.

EYE

• If this product comes in contact with the eyes: · Immediately hold eyelids apart and flush the eye continuously with running water. · Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.

SKIN

■ If skin or hair contact occurs: · Immediately flush body and clothes with large amounts of water, using safety shower if available. · Quickly remove all contaminated clothing, including footwear.

#### INHALED

· If fumes or combustion products are inhaled remove from contaminated area. · Lay patient down. Keep warm and rested.

#### NOTES TO PHYSICIAN

#### Treat symptomatically.

for poisons (where specific treatment regime is absent):

-----BASIC TREATMENT

· Establish a patent airway with suction where necessary.

· Watch for signs of respiratory insufficiency and assist ventilation as necessary.

Material may produce delayed skin burns and these may be second degree burns with little or no warning.

Liver and kidney function should be followed after a significant exposure.

Lung function should be assessed following a significant single or intermittent inhalation exposure.

If burn is present, treat as any thermal burn, after decontamination.

No special antidote.

Supportive care. Treatment based on judgment of the physician in response to reactions of the patient.

Section 5 - FIRE FIGHTING MEASURES				
Vapour Pressure (mmHG):	Not available			
Upper Explosive Limit (%):	Not available			
Specific Gravity (water=1):	1.601			
Lower Explosive Limit (%):	Not available			

#### **EXTINGUISHING MEDIA**

· Water spray or fog.

· Foam.

#### FIRE FIGHTING

· Alert Emergency Responders and tell them location and nature of hazard.

· Wear breathing apparatus plus protective gloves for fire only.

When any large container (including road and rail tankers) is involved in a fire,

consider evacuation by 1000 metres in all directions.

## GENERAL FIRE HAZARDS/HAZARDOUS COMBUSTIBLE PRODUCTS

· Liquid and vapor are flammable.

 $\cdot$  Moderate fire hazard when exposed to heat or flame.

Combustion products include: carbon monoxide (CO), carbon dioxide (CO2), other pyrolysis products typical of burning organic material. May emit poisonous fumes.

## FIRE INCOMPATIBILITY

Avoid contamination with oxidizing agents i.e. nitrates, oxidizing acids, chlorine bleaches, pool chlorine etc. as ignition may result.

## PERSONAL PROTECTION

Glasses: Full face- shield. Gloves: 1.BUTYL 2.PVA 3.VITON Respirator: Type A Filter of sufficient capacity

# Section 6 - ACCIDENTAL RELEASE MEASURES

#### MINOR SPILLS

- · Remove all ignition sources.
- $\cdot$  Clean up all spills immediately.

MAJOR SPILLS

- · Clear area of personnel and move upwind.
- $\cdot$  Alert Emergency Responders and tell them location and nature of hazard.

# Section 7 - HANDLING AND STORAGE

## PROCEDURE FOR HANDLING

 $\cdot$  DO NOT allow clothing wet with material to stay in contact with skin.

- · Avoid all personal contact, including inhalation.
- · Wear protective clothing when risk of exposure occurs.
- · Containers, even those that have been emptied, may contain explosive vapours.
- $\cdot$  Do NOT cut, drill, grind, weld or perform similar operations on or near containers.

#### **RECOMMENDED STORAGE METHODS**

- · Lined metal can, Lined metal pail/drum
- · Plastic pail.
- For low viscosity materials
- · Drums and jerricans must be of the non-removable head type.

· Where a can is to be used as an inner package, the can must have a screwed enclosure.

All inner and sole packagings for substances that have been assigned to Packaging Groups I or II on the basis of inhalation toxicity criteria, must be hermetically sealed.

#### STORAGE REQUIREMENTS

- Store in original containers.
- · Keep containers securely sealed.

# Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

#### **EXPOSURE CONTROLS**

Source	Material	TWA ppm	TWA mg/m³	STEL ppm	STEL mg/m³	Peak ppm	Peak mg/m³	TWA F/CC	Notes
<u> </u>									
US TSCA New Chemical Exposure Limits (NCEL)	epibromohydrin (Halogenated alkanes (P84-106/107))	1.0							

ENDOELTABLE

#### PERSONAL PROTECTION



#### RESPIRATOR

Type A Filter of sufficient capacity

Consult your EHS staff for recommendations

EYE

· Chemical goggles.

# · Full face shield.

# HANDS/FEET

. When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.

Elbow length PVC gloves.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include: such as:

- · frequency and duration of contact,
- · chemical resistance of glove material,
- glove thickness and
- · dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739).

• When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374) is recommended.

· When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374) is recommended.

· Contaminated gloves should be replaced.

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

#### OTHER

Overalls.

· Evewash unit.

· Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity.

· For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets), non sparking safety footwear.

#### **ENGINEERING CONTROLS**

■ Local exhaust ventilation usually required. If risk of overexposure exists, wear an approved respirator.

# Section 9 - PHYSICAL AND CHEMICAL PROPERTIES

### PHYSICAL PROPERTIES

Liquid. Toxic or noxious vapours/gas.			
State	Liquid	Molecular Weight	136.98
Melting Range (°F)	-40	Viscosity	Not available
Boiling Range (°F)	273.2- 276.8	Solubility in water (g/L)	Reacts
Flash Point (°F)	132.8	pH (1% solution)	Not applicable
Decomposition Temp (°F)	Not available	pH (as supplied)	Not applicable
Autoignition Temp (°F)	Not available	Vapour Pressure (mmHG)	Not available
Upper Explosive Limit (%)	Not available	Specific Gravity (water=1)	1.601
Lower Explosive Limit (%)	Not available	Relative Vapor Density (air=1)	>1
Volatile Component (%vol)	Not available	Evaporation Rate	Not available

### APPEARANCE

Colourless liquid; reacts with water.

Experimental and modelled log Kow values of 0.68 and 0.86, respectively, indicate that the potential for bioaccumulation of ethyloxirane in organisms is likely to be low. Modelled bioaccumulation -factor (BAF) and bioconcentration -factor (BCF) values of 1 to 17 L/kg indicate that ethyloxirane does not meet the bioaccumulation criteria (BCF/BAF = 5000)\*

Material

Value

# Section 10 - CHEMICAL STABILITY

# CONDITIONS CONTRIBUTING TO INSTABILITY

 $\cdot$  Presence of incompatible materials.

· Product is considered stable.

#### STORAGE INCOMPATIBILITY

· Epoxides are highly reactive with acids, bases, and oxidizing and reducing agents.

- · Epoxides react, possibly with anhydrous metal chlorides, ammonia, amines and group 1 metals.
- Segregate from alcohol, water.
- · Avoid strong acids, bases.

Avoid reaction with oxidizing agents.

· NOTE: May develop pressure in containers; open carefully. Vent periodically.

Epichlorohydrin:

• may polymerise explosively at higher temperatures, in the presence of strong acids, strong bases, metallic halides, aluminium, aluminium chloride, iron(III) chloride, or zinc

· reacts violently with aniline, hypochlorite, isopropylamine, potassium tert-butoxide, sulfuric acid

· mixtures with trichloroethylene form dichloroacetylene, an explosive

· is incompatible with aliphatic amines, alkaline earths, alkali metals, alkanolamines, powdered metals, strong oxidisers - fire and explosion may result

· decomposition produces highly toxic phosgene gas

- · will pit steel in the presence of moisture
- · flow or agitation may generate electrostatic charges dues to low conductivity.

Avoid zinc, sodium, aluminium, magnesium and their alloys.

For incompatible materials - refer to Section 7 - Handling and Storage.

# Section 11 - TOXICOLOGICAL INFORMATION

#### epibromohydrin

#### TOXICITY AND IRRITATION

EPIBROMOHYDRIN:

unless otherwise specified data extracted from RTECS - Register of Toxic Effects of Chemical Substances.
TOXICITY IRRITATION

Intraperitoneal (rat) LD50: 300 mg/kg Nil Reported

■ Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of

minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production.

Oxiranes (including glycidyl ethers and alkyl oxides, and epoxides) exhibit many common characteristics with respect to animal toxicology. One such oxirane is ethyloxirane; data presented here may be taken as representative.

### for 1,2-butylene oxide (ethyloxirane):

Ethyloxirane increased the incidence of tumours of the respiratory system in male and female rats exposed via inhalation. Significant increases in nasal papillary adenomas and combined alveolar/bronchiolar adenomas and carcinomas were observed in male rats exposed to 1200 mg/m3 ethyloxirane via inhalation for 103 weeks. There was also a significant positive trend in the incidence of combined alveolar/bronchiolar adenomas and carcinomas. Nasal papillary adenomas were also observed in 2/50 high-dose female rats with none occurring in control or low-dose animals. In mice exposed chronically via inhalation, one male mouse developed a squamous cell papilloma in the nasal cavity (300 mg/m3) but other tumours were not observed. Tumours were not observed in mice exposed chronically via dermal exposure. When trichloroethylene containing 0.8% ethyloxirane was administered orally to mice for up to 35 weeks, followed by 0.4% from weeks 40 to 69, squamous-cell carcinomas of the forestomach occurred in 3/49 males (p=0.029, age-adjusted) and 1/48 females at week 106. Trichloroethylene administered alone did not induce these tumours and they were not observed in control animals . Two structurally related substances, oxirane (ethylene oxide) and methyloxirane (propylene oxide), which are also direct-acting alkylating agents, have been classified as carcinogenic.

#### CARCINOGEN

BROMINE COMPOUNDS (ORGANIC	US Environmental Defense Scorecard	Deference(a)	P65-MC
OR INORGANIC)	Suspected Carcinogens	Relefence(s)	P05-IVIC

# Section 12 - ECOLOGICAL INFORMATION

Harmful to aquatic organisms.

This material and its container must be disposed of as hazardous waste.

#### Ecotoxicity

Ingredient epibromohydrin Persistence: Water/SoilPersistence: AirBioaccumulationMobilityHIGHLOWHIGH

# Section 13 - DISPOSAL CONSIDERATIONS

#### **US EPA Waste Number & Descriptions**

A. General Product Information

Ignitability characteristic: use EPA hazardous waste number D001 (waste code I)

#### **Disposal Instructions**

All waste must be handled in accordance with local, state and federal regulations.

Puncture containers to prevent re-use and bury at an authorized landfill.

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- · Reduction
- · Reuse
- Recycling
- · Disposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.

DO NOT allow wash water from cleaning equipment to enter drains. Collect all wash water for treatment before disposal.

· Recycle wherever possible.

· Consult manufacturer for recycling options or consult Waste Management Authority for disposal if no suitable treatment or disposal facility can be identified.

# Section 14 - TRANSPORTATION INFORMATION

DOT:

Symbols: None Hazard class or Division: 6.1 Identification Numbers: UN2558 PG: I Label Codes: 6.1, 3 Special provisions: T14, TP2, TP13 Packaging: Exceptions: None Packaging: Non- bulk: 201 Packaging: Exceptions: None Quantity limitations: Forbidden Passenger aircraft/rail: Quantity Limitations: Cargo Forbidden Vessel stowage: Location: D aircraft only: Vessel stowage: Other: 40 S.M.P.: YES

Hazardous materials descriptions and proper shipping names:

#### Epibromohydrin Air Transport IATA:

Air Transport IATA: ICAO/IATA Class: 6.1 ICAO/IATA Subrisk: None UN/ID Number: 2558 Packing Group: -Special provisions: None Cargo Only Packing Instructions: Forbidden Maximum Qty/Pack: Forbidden Passenger and Cargo Passenger and Cargo Packing Instructions: Forbidden Maximum Qty/Pack: Forbidden Passenger and Cargo Limited Quantity Passenger and Cargo Limited Quantity Packing Instructions: Forbidden Maximum Qty/Pack: Forbidden Passenger and Cargo Limited Quantity Passenger and Cargo Limited Quantity Packing Instructions: Forbidden Maximum Qty/Pack: Forbidden Shipping Name: EPIBROMOHYDRIN

# Maritime Transport IMDG:

IMDG Class: 6.1 IMDG Subrisk: 3-P UN Number: 2558 Packing Group: I EMS Number: F-E , S-D Special provisions: None Limited Quantities: 0 Shipping Name: EPIBROMOHYDRIN

# Section 15 - REGULATORY INFORMATION

# epibromohydrin (CAS: 3132-64-7) is found on the following regulatory lists;

"Canada Ingredient Disclosure List (SOR/88-64)","Canada Non-Domestic Substances List (NDSL)","Canada Toxicological Index Service -Workplace Hazardous Materials Information System - WHMIS (English)","US - New Jersey Right to Know Hazardous Substances","US Department of Transportation (DOT) Marine Pollutants - Appendix B","US DOE Temporary Emergency Exposure Limits (TEELs)","US Toxic Substances Control Act (TSCA) - Inventory","US TSCA Section 12(b) - List of Chemical Substances Subject to Export Notification Requirements", "US TSCA Section 5(a)(2) - Significant New Use Rules (SNURs)"

# Section 16 - OTHER INFORMATION

#### ND

Substance CAS Suggested codes epibromohydrin 3132- 64- 7 AUTOID~

Reasonable care has been taken in the preparation of this information, but the author makes no warranty of merchantability or any other warranty, expressed or implied, with respect to this information. The author makes no representations and assumes no liability for any direct, incidental or consequential damages resulting from its use. For additional technical information please call our toxicology department on +800 CHEMCALL.

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references. A list of reference resources used to assist the committee may be found at: www.chemwatch.net/references.

• The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

This document is copyright. Apart from any fair dealing for the purposes of private study, research, review or criticism, as permitted under the Copyright Act, no part may be reproduced by any process without written permission from CHEMWATCH. TEL (+61 3) 9572 4700.

Issue Date: Oct-4-2009 Print Date:Mar-4-2011